Magnetic Properties of Transition Metal Complexes from Density Functional Theory

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Outline

- Density Functional Theory in a Nutshell
- Molecular Magnets
- Magnetic Properties
- Fe₇ disks



Electrons and nuclei: Materials and Chemistry

Born-Oppenheimer approximation

Holy grail

 $E, \psi(\mathbf{r}_1, \mathbf{r}_2, \mathbf{r}_3, \dots, \mathbf{r}_n)$

$$H = \sum_{i}^{n} (T_{i} - \sum_{N} \frac{Z_{N}}{\left|\vec{r}_{i} - \vec{R}_{N}\right|}) + \sum_{i,j}^{n} \frac{1}{r_{ij}}$$



 $\psi(\mathbf{r}_1,\mathbf{r}_2,\mathbf{r}_3,...,\mathbf{r}_n)$

Extremely difficult to solve in practice:

- Must include spin: \otimes SU(2).
- 3*n* electronic coordinates. Complexity grows exponentially with that number.
- Solutions need to be antisymmetric with respect to particle exchange.



"the difficulty is only that the exact application of these laws leas to equations much too complicated to be soluble"

Hohenberg and Kohn (1964)

$$E = E[n(\vec{r})]$$

Energy is a functional of the one-electron density!







Now the problem reduces to 3 degrees of freedom!

Computational complexity: N^{α}



Kohn and Sham (1965)

$$n(\vec{r}) = \sum_{i \in occ} \left| \varphi_i(\vec{r}) \right|^2$$

and

$$E = E_{ext} + T + E_{coul} + E_{xc}$$

We know how to calculate these

Unknown

(Nobel prize in Chemistry in 1998)

Bad news:

The functional E[n] is unknown...



After more than 50 years and trying very hard....

...we have to use approximations

 $E = E_{ext} + T + E_{coul} + E_{xc}[n, \nabla n, \nabla^2 n, etc.]$ Density (LDA) Density gradient (GGA) Lapacian of the density (meta-GGA) Kinetic energy density (meta-GGA) Hartree-Fock exchange (hybrid)

⊕Exact theory, in principle.
⊕Computationally easy to implement.
⊕Fast!

Some widely used functionals:

- > LDA: The first and simplest approximation (1980)
- GGAs : PW91, PBE, very successful in solid state (1991, 1997)
- Hybrid functionals (Becke, 1993): popular in chemistry: B3LYP

DFT is a theory for the ground state only!

Density Functional Theory in Practice



Density Functional Theory in Practice

- ⇒ No systematic (in the mathematical sense) way of approaching the exact functional.
- \Rightarrow Known exact properties can be used.
- ⇒ Comparison:
 - ⇒ Very accurate numerical solutions.
 - ⇒ Very accurate experiments.

Typical mean absolute errors:

Dissociation energies	~4 kcal/mol	B3LYP
Harmonic frequencies	~ 30 cm ⁻¹	VSXV
Bond lengths	~0.01Å	PBE hybrid
Lattice constants	~0.04Å	LDA
NMR ¹³ C shieldings	~3 ppm	B3LYP

Ideally, we would like to use the one functional for all properties

However, functionals perform different for different properties

Many spectacular failures and successes

Molecular Magnets

- Single molecule magnets (or molecular magnets) are a small group of paramagnetic metal ions bound together by simple groups like O²⁻, OH⁻, Cl⁻, OCH₃⁻, etc.
- First SMM reported: Mn₁₂ complex; Sessoli et al., Nature 365, 141 (1993).



- Organic ligands keep the structure in place and shield the metal ions.
- The ions are typically 3*d* transition metals.
- The ions are magnetically coupled (superexchange).

Postnikov, Kortus, and Pederson, Phys. Stat. Sol. (b) 243, 2533 (2006)

Molecular Magnets

- These molecules show slow relaxation of the magnetization.
- As a consequence, it "remembers" its magnetization after a magnetic field is switched off, just like traditional magnets.



• This is a purely molecular property, unlike macroscopic magnets.



Wikipedia; Postnikov, Kortus, and Pederson, Phys. Stat. Sol. (b) 243, 2533 (2006)

Applications:

- These are the smallest magnets that can be made: High density data storage.
- Spintronics: building devices that use the spin current to operate.
- Quantum computing: The spin state of a MM can be used as a qbit.



Mount Pleasant (Michigan)



- Science of Advanced Materials PhD
- Physics + Chemistry + Engineering

- Multidisciplinary program.
- Strong in electronic structure.
- Nice, quiet town.

Looking for enthusiastic students!

How can we model the magnetic behavior of a SMM?

 Model spin Hamiltonians (Heisenberg-Dirac-van Vleck) in most cases capture the physics

$$\widehat{H} = -2\sum_{\langle AB \rangle} J_{AB} \widehat{\mathbf{S}}_A \cdot \widehat{\mathbf{S}}_B$$

- <u>Quantum</u> or classical
- Used to model a large number of problems in physics :
 - Spin waves, Ising model, etc.
- "Magnetism in Condensed Matter", S. Blundell (Oxford, 2001)





Simple model Hamiltonians can reproduce a huge range of scales!

From W. Wernsdorfer, Int. J. Nanotechnology

 The success of this model is related to the ability to model the energetics of a large number of low-lying states.

$$H = -2\sum_{\langle AB \rangle} J_{AB} \, \mathbf{S}_A \cdot \mathbf{S}_B$$

"Magnetism in Condensed Matter", S. Blundell (Oxford, 2001)



Example: Simple s=1/2 dimer



Zero field splitting:

- Magnetic Anisotropy Energy (spin-orbit)
- Magnetodipolar interactions (small ~0.1 cm⁻¹)

$$\widehat{H}_{HDVV} = -J \sum_{\langle ij \rangle} \widehat{S}_i \cdot \widehat{S}_j + \widehat{S} \cdot \mathbf{D} \cdot \widehat{S}$$
3x3 tensor Total molecular spin

$$\widehat{H}_{HDVV} = -J \sum_{\langle ij \rangle} \widehat{\mathbf{S}}_i \cdot \widehat{\mathbf{S}}_j + E(\widehat{\mathbf{S}}_x^2 - \widehat{\mathbf{S}}_y^2) + D(\widehat{\mathbf{S}}_z^2 - S(S+1)/3)$$
$$\Delta M = 1 \text{ or } \Delta M = 2$$

With
$$D = \frac{3D_{ZZ}}{2}$$
 and $E = \frac{|D_{XX} - D_{YY}|}{2}$



Real complexes:

- Exchange interactions are generally negative ~0-100 cm^{-1.}
- Spin-orbit is responsible for the magnetic anisotropy.
- Orbital angular momentum does not contribute to magnetization (usually due to spin-orbit and Jahn-Teller distortion).
- Tens to hundreds atoms.
- Several transition metal centers:
 - Fe₈: 8 s=5/2 centers \rightarrow ~10⁶ configurations
 - Mn₁₂: 4 s=3/2 plus 8 s=2 centers → ~10⁸ configurations
- Thermal spin inversion originates in the interactions with the environment: phonons.



 $U_{eff}{\sim}S^2|D|$

For Mn_{12}

- *S* = 10
- $D = -0.5 \text{ cm}^{-1}$
- $U_{eff} \sim 50 \text{ cm}^{-1} \text{ or } 60 \text{ K}$

L. Bogani and W. Wernsdorfer, Nature mater. 7, 179 (2008)

Experiments

Example: Asymmetric [Mn₃Ca₂] Mukherjee et al., PNAS 109, 2257 (2012).

- S_T determined from variable T DC and AC magnetic susceptibility.
- 2-J model: 64 states, 12 unique energies.
- Fits experimental susceptibility very well with $S_T=9/2$ and J=40 cm⁻¹ J'=-11 cm⁻¹.



Experiments

Example: Asymmetric [Mn₃Ca₂] Mukherjee et al., PNAS 109, 2257 (2012).

• 2-J model: 64 states, 12 unique energies.



- For large molecules with many Js, fitting becomes difficult
- Fitting is more difficult for FM coupling

Real complexes:



Quantum states of Fe₈

- Clearly very difficult for electronic structure.
- What kind of information can we get from DFT?
 - Parameters in the HDVV Hamiltonian: *J* couplings and MAE

Delfs et al., Inorganic Chemistry 32, 199 (1993)

Magnetic Anisotropy Energy

- Spin-orbit interaction (fine structure in atoms):
- For 3d elements it can be included as a perturbation.
 - Consider second-order perturbation theory with different directions for the spin of the reference state.
- For heavier elements (such as Lanthanides) perturbation theory is not a good approximation.
 - Fully self-consistent with spin-orbit: 2-components or "noncollinear" DFT.
 - Relativistic approaches: 4-components.
- The D parameter is affected by the "local" distortions. See Cirera et al., Chem. Eur. J. 15, 4078 (2009).



Neese and Solomon, Inorg. Chem. **37**, 6568 (1998) Pederson and Khanna, Phys. Rev. B **59**, R693 (1999) van Wüllen, J. Chem. Phys. **130**, 194109 (2009)

Molecular Magnets

Table 1. Calculated D values $[cm^{-1}]$ for mononuclear complexes obtained by using the PBE functional and NRLMOL code that only include the spin-orbit term. D values were also calculated with the ORCA code and included the spin-spin contributions. Experimental data obtained by EPR techniques are provided for comparison.

2		d ⁿ	$D_{\text{calcd(SO)}}$	$D_{\text{calcd}(\text{SO+SS})}$	$D_{\rm exptl}$	Ref.
[Mn(dbm) ₃]	Mn ^{III}	d ⁴	-2.43	-3.49	-4.57	[43]
[Mn(acac) ₃]	Mn ^{III}	d ⁴	-2.36	-3.28	-4.52 ^[a]	[12]
[MnF ₃ (terpy)]	Mn ^{III}	d ⁴	-1.84	-2.65	-3.82 ^[a]	[44]
[MnF ₃ (bpea)]	Mn ^{III}	d ⁴	-1.84	-2.63	-3.67 ^[a]	[44]
$[Mn(N_3)_3(terpy)]$	Mn ^{III}	d ⁴	-1.65	-2.39	-3.29 ^[a]	[45]
[MnCl(Py)(TPP)]	Mn ^{III}	d ⁴	-1.28	-1.93	-3.00 ^[a]	[46]
[Fe(dpm) ₃]	Fe ^{III}	d ⁵	-0.19	+0.26	-0.18	[47]
[Fe(mal) ₃]	Fe ^{III}	d ⁵	-0.13	-0.11	± 0.12	[48]
[Fe(acac) ₃]	Fe ^{III}	d ⁵	+0.29	-0.22	+0.16	[48]
$[Mn(bpea)(N_3)_3]$	Mn ^{III}	d^4	+1.41	+2.07	$+3.50^{[a]}$	[44]
$[Fe(SPh)_4]^{2-}$	Fe ^{II}	d ⁶	+2.18	+3.14	+5.48 ^[a]	[49]

[a] Values obtained from high-field EPR spectroscopy.

phys. stat. sol. (b) 243, No. 11 (2006)

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 Table 3
 Comparison of the calculated by NRLMOL and experimental magnetic anisotropy parameter D for the single molecule magnets. See theory references for computational details.

molecule	S	<i>D</i> (K)	
		theory	experiment
$Mn_{12}O_{12}(O_2CH)_{16}(H_2O)_4$	10	-0.56^{a}	-0.56 ^b
$[Fe_8O_2(OH)_{12}(C_6H_{15}N_3)_6Br_6]^{2+}$	10	-0.53°	-0.30^{d}
$[Mn_{10}O_4(2,2'-biphenoxide)_4Br_{12}]^{4-}$	13	-0.06^{e}	-0.05^{f}
Co ₄ (CH ₂ C ₅ H ₄ N) ₄ (CH ₃ OH) ₄ Acl ₄	6	-0.64^{g}	$-0.70.9^{h}$
Fe ₄ (OCH ₂) ₆ (C ₄ H ₉ ON) ₆	5	-0.56^{i}	-0.57^{j}
$Cr[N(Si(CH_3)_3)_2]_3$	3/2	-1.15^{i}	-2.66^{k}
Mn ₉ O ₃₄ C ₃₂ N ₃ H ₃₅	17/2	-0.33^{m}	-0.32^{1}
Ni ₄ O ₁₆ C ₁₆ H ₄₀	4	-0.385	-0.40^{m}
$Mn_4O_3Cl_4(O_2CCH_2CH_3)_3(NC_5H_5)_3$	9/2	-0.58^{n}	-0.72°

Magnetic Exchange Couplings

• Energy differences methods: The different states of the HDVV Hamiltonian are "mapped" into DFT solutions



- It provides a simple scheme to extract J couplings
- It can be easily extended to other cases
- Widely used in the literature

Problems with the energy differences method

- Multicenter TM complexes can have several target states and convergence to these can be difficult.
- Cannot be done as a "black-box".

Can we extract J couplings in a more efficient way γ

Consider local spin rotations





- Implemented as a constraint using Lagrange multipliers
- Uses analytical linear response
- All couplings can be extracted from a single reference state
- Close to energy differences for "Heisenberg" like systems
- Works for multinuclear and heteronuclear complexes
- Needs noncollinear kernel



 $W = E^{DFT} - \lambda_{AB}(Q_{AB} - \theta_{AB}) \text{ where } \lambda_{AB} \text{ represents is a Lagrange multiplier}$



All we need is $\frac{\partial^2 E}{\partial \lambda_{AB} \partial \lambda_{CD}}$ that can be calculated using perturbation theory.

Other approaches:

- Liechtenstein et al., J. Phys. F 14, L125 (1984)
- Bruno, PRL 90, 087205 (2003)
- Zhekova et al., JCTC **7**, 1858 (2011)



	LSDA	+15%HFX	+30%HFX	Exp.
J ₁₂	-102	-41	-21	-21
J_{13}	-96	-41	-21	-21
J_{14}	-103	-44	-22	-21
J ₂₃	-2	-1	~0	small
J ₂₄	-3	-2	~0	small
J ₃₄	-3	0	~0	small

J couplings: Performance of DFT



Phillips and Peralta, J. Chem. Phys. 134, 034108, (2011)

Assessment in a set of 16 bimetallic TM complexes with accurate experimental reference values



Fe(III)₇ disks: Mukherjee et al., Inorg. Chem. 50, 3849 (2011)



S=5/2 ground state

High-spin ground state

- Fe(III) typically exhibits antiferromagnetic coupling
- Too many Fe centers to fit susceptibility data: Couplings are very difficult/impossible to determine experimentally

Complex 1

Couplings in cm⁻¹



From magneto-structural data [Mukherjee et al., Inorg. Chem. **50**, 3849 (2011)]

Alternated strong and weak couplings





Complex 2

Complex 3

Strong AF "radial" couplings



Alternated coupling dominates



Summary

- DFT is a powerful tool for materials science, chemistry, and physics.
- It provides fast, sometimes accurate computational results.
- This is also the case for magnetic properties, although more work needs to be done.

Acknowledgements









Questions? Comments?