

Joop van Lenthe



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t           t           l
tttt       u  u       rrrrr  tttt       l           eee
t           u  u       r   r   t           l           eeeee
t   t     u  u       r           t   t     l   l     e
  ttt      uu       r           ttt      lll     eee

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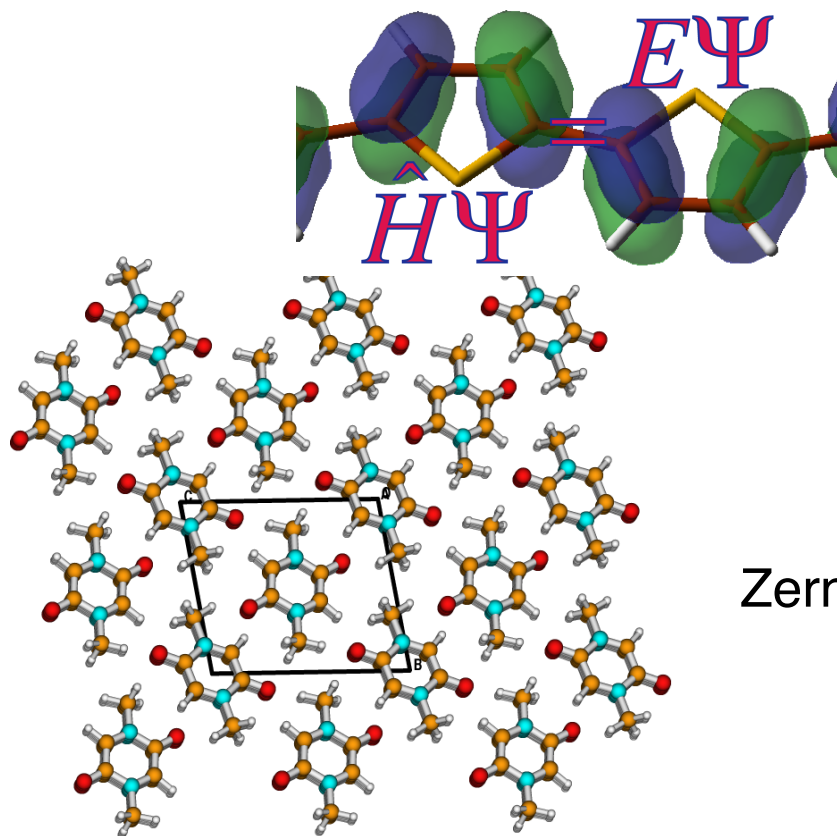
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Non orthogonal configuration interaction



*Remco W. A. Havenith
and
Ria Broer*

Theoretical Chemistry,
Zernike Institute for Advanced Materials,
Stratingh Institute for Chemistry,
University of Groningen,
The Netherlands

Ghent Quantum Chemistry Group, Ghent University, Belgium



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Theoretical Chemistry

Team



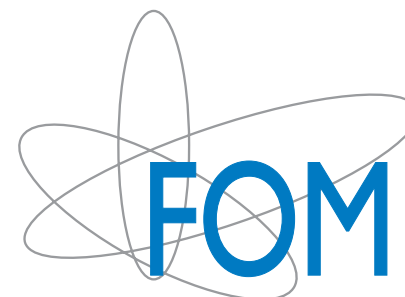
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Ria Broer



Meilani Wibowo



Kathir Kanagarajan



Coen de Graaf



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Tjerk Straatsma



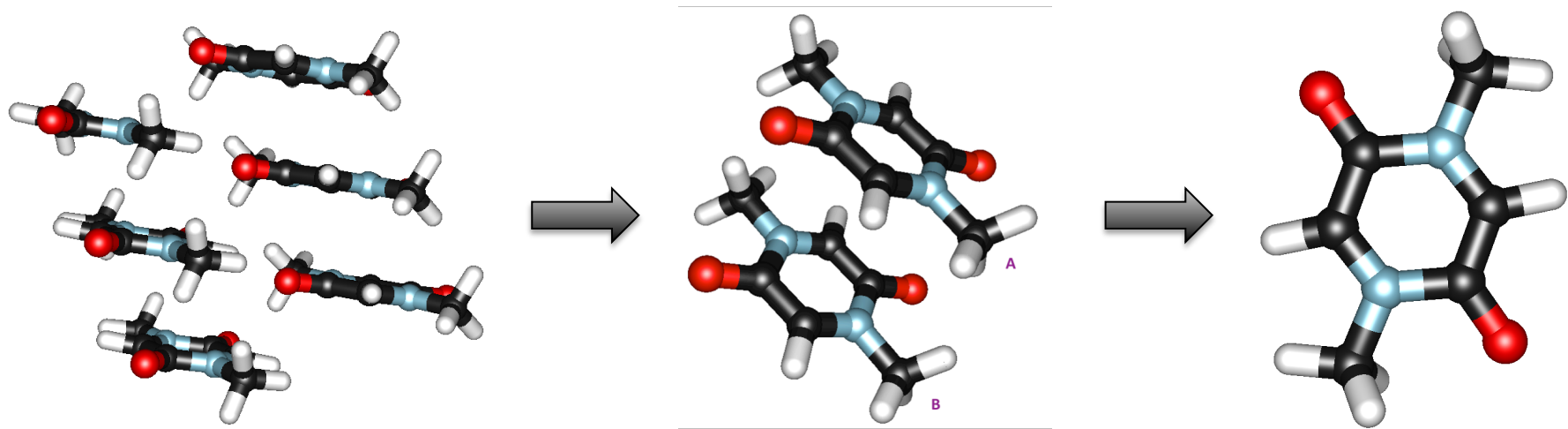
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Theoretical Chemistry

Cluster approximation for solids



- Describe solid in terms of molecular wavefunctions
- Compute wavefunctions of each molecule for specific states (CASSCF)
- Form many-electron basis functions (S_0S_0 , S_0S_1 , 1TT , CT), each describing a particular combination of molecular states

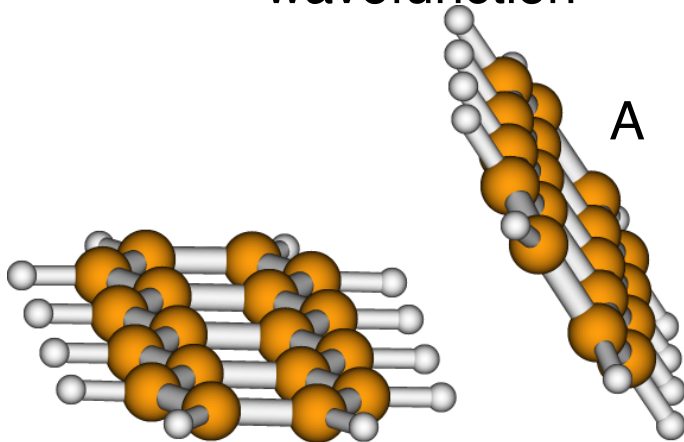


Non orthogonal configuration interaction

- Describe wavefunctions of a cluster of molecules in terms of (localised) molecular many-electron basis functions (MEBF)
 - MEBFs are antisymmetrised products of molecular wavefunctions:

$$\Phi_{AB}^{KL} = A(\Phi_A^K \times \Phi_B^L)$$

- Molecular wavefunction can be any multiconfigurational wavefunction



Ψ^0 : Ground state

Ψ^1 : Singlet excited state

Ψ^T : Triplet excited state

$$\Phi_{AB}^{0 \times 0} = A(\Psi_A^0 \times \Psi_B^0)$$

$$\Phi_{AB}^{1 \times 0} = A(\Psi_A^1 \times \Psi_B^0)$$

$$\Phi_{AB}^{0 \times 1} = A(\Psi_A^0 \times \Psi_B^1)$$

$$\Phi_{AB}^{T \times T} = A(\Psi_A^T \times \Psi_B^T)$$

⋮

B

$$\Psi = c_1 \Phi_{AB}^{0 \times 0} + c_2 \Phi_{AB}^{1 \times 0} + c_3 \Phi_{AB}^{0 \times 1} + c_4 \Phi_{AB}^{T \times T} + \dots$$



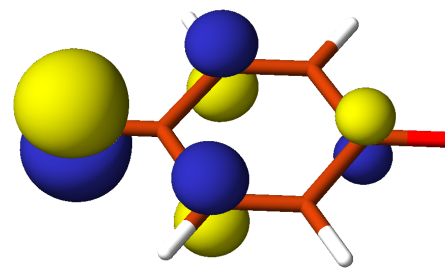
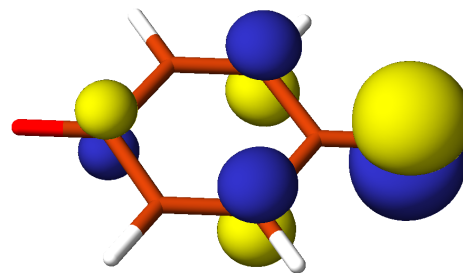
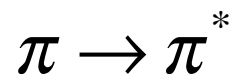
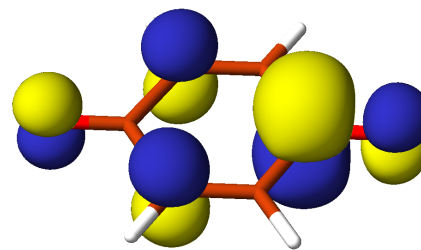
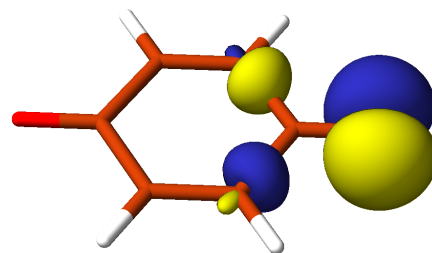
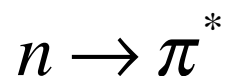
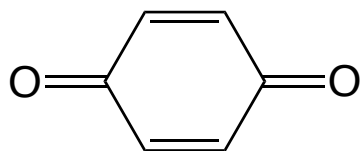
Non Orthogonal Configuration Interaction

- Wavefunction expanded as: $\Psi = \sum_{i=1}^N c_i \Phi_i$ with Φ_i a many-electron basis function ((MEBF) Slater determinant, or combination thereof)
- The orbitals χ_j in a MEBF are not orthogonal, making the many-electron MEBFs also not orthogonal: $\langle \Phi_i | \Phi_j \rangle = S_{ij}$
- The non orthogonality of the orbitals within one MEBF and of the orbitals in a different MEBF complicates the calculation of the required Hamiltonian matrix elements $\langle \Phi_i | H | \Phi_j \rangle$
- Solve $(\mathbf{H} - E\mathbf{S})(\mathbf{c}) = 0$ to get energies and Ψ (c_i 's)



Advantages of this NOCI

- Inclusion of orbital relaxation effects
- Inclusion of (static) correlation effects
- Short wavefunction expansions
- Chemical interpretability
 - Description of system in terms of predefined states



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Theoretical Chemistry

Computational Aspects of our NOCI approach

- Typical number of $|\Phi_{AB}^{KL}\rangle \sim 20$
 - H/S matrices contain ~ 210 elements of the type
 $\langle \Phi_{AB}^{KL} | H | \Phi_{AB}^{K'L'} \rangle$ and $\langle \Phi_{AB}^{KL} | \Phi_{AB}^{K'L'} \rangle$
- If $|\Phi_A^K\rangle$ contains ~ 500 determinants, then $|\Phi_A^K\rangle \times |\Phi_B^L\rangle = |\Phi_{AB}^{KL}\rangle \sim 2.5 \cdot 10^5$
- $\langle \Phi_{AB}^{KL} | H | \Phi_{AB}^{K'L'} \rangle = \sum_i \sum_j c_i c_j \langle \Delta_i | H | \Delta_j \rangle$
- Approximately 10^7 elements $\langle \Delta_i | H | \Delta_j \rangle$ have to be calculated for one matrix element $\langle \Phi_{AB}^{KL} | H | \Phi_{AB}^{K'L'} \rangle$
- Aim for high level of parallelism
- Easy to parallelize



Current program

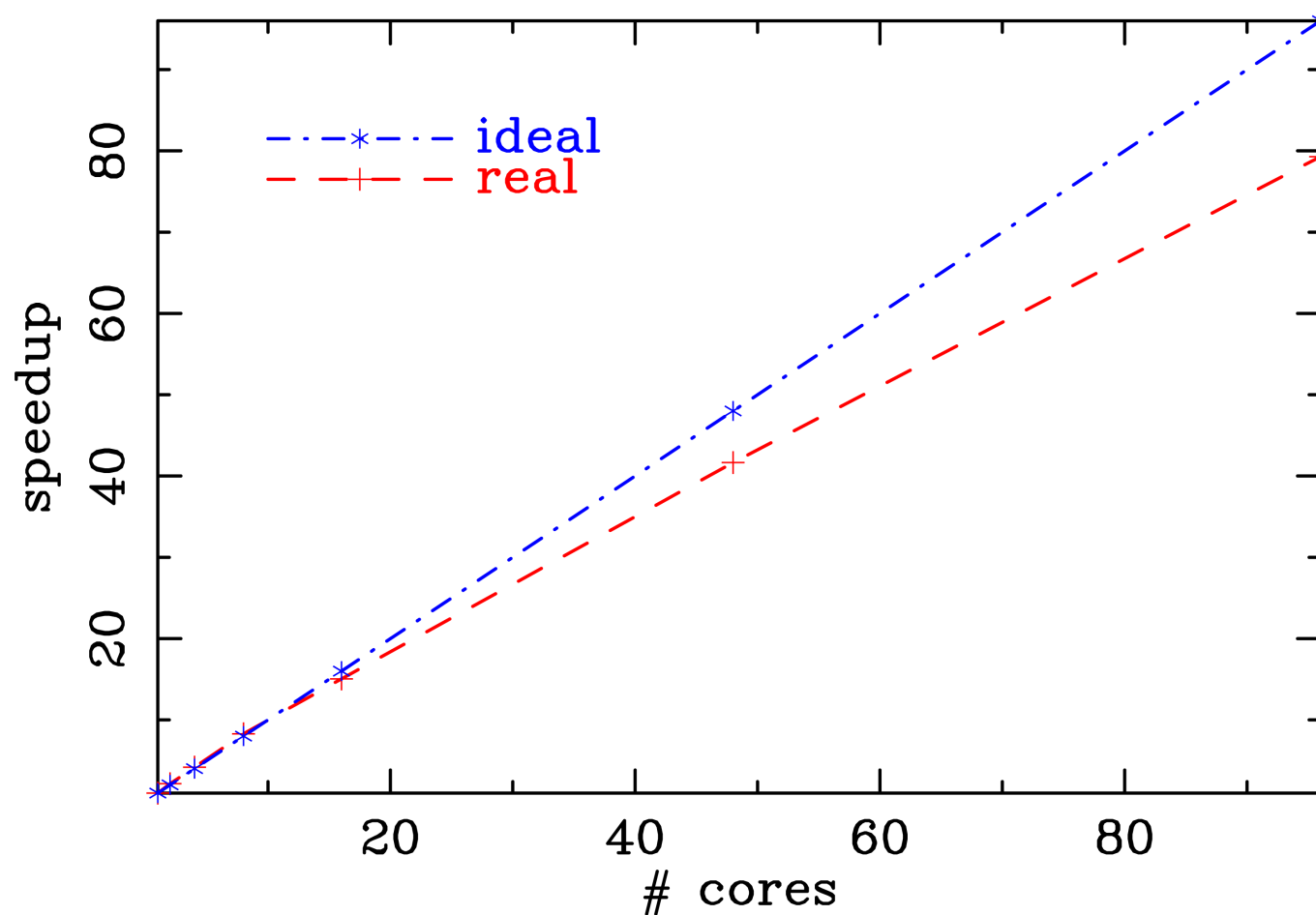
- Based on existing GNOME code for evaluation of $\langle \Delta_i | H | \Delta_j \rangle$
- First, remove all i, j combinations whose occupations differ by more than two electrons, and all combinations for which $|c_i c_j| < \varepsilon$
- Parallelized using MPI and OpenMP
- Evaluation of $\langle \Delta_i | H | \Delta_j \rangle$ in parallel
 - All elements are divided over the nodes, then divided over

```
the cores    !$OMP DO SCHEDULE(DYNAMIC) REDUCTION(+:hmat)
             do ij=ijstart,ijend
               i=iloop(ij,1)
               j=iloop(ij,2)
               call gnome(hh,ss)
               hmat=hmat+hh
             enddo
             !$OMP END PARALLEL
```



Performance

- Test case I: 52 determinants, 1378 matrix elements $\langle \Delta_i | H | \Delta_j \rangle$



Technical Aspects

- Evaluation of $\langle \Delta_i | H | \Delta_j \rangle$ with non orthogonal orbitals
 - $H_{ij} = \sum_{i,j} h_{ij} S^{(i,j)} + \sum_{i < k} \sum_{j < l} [(ij | kl) - (ik | jl)] S^{(i,j,k,l)}$
 - First and second order co-factors needed
- With corresponding orbitals, then $\langle c_i | d_j \rangle = \lambda_i \delta_{ij}$
and $S^{(i,i)} = \prod_{m \neq i} \lambda_m$ ($S^{(i,j)} = 0$ for $i \neq j$)
- No 4-index, but transform co-factors to common AO basis in which the corresponding orbitals c_i and d_i are expressed
- SVD and matrix multiplications
- Use GPUs?
- Use molecular orbitals as one-electron basis



The procedure - NOCI

OLD approach (MOLCAS)

- CASCI wavefunction as approximation to ASP
 - Mixing in contaminating CSFs
 - (Unwanted orthogonalizations)

NEW approach (GNOME)

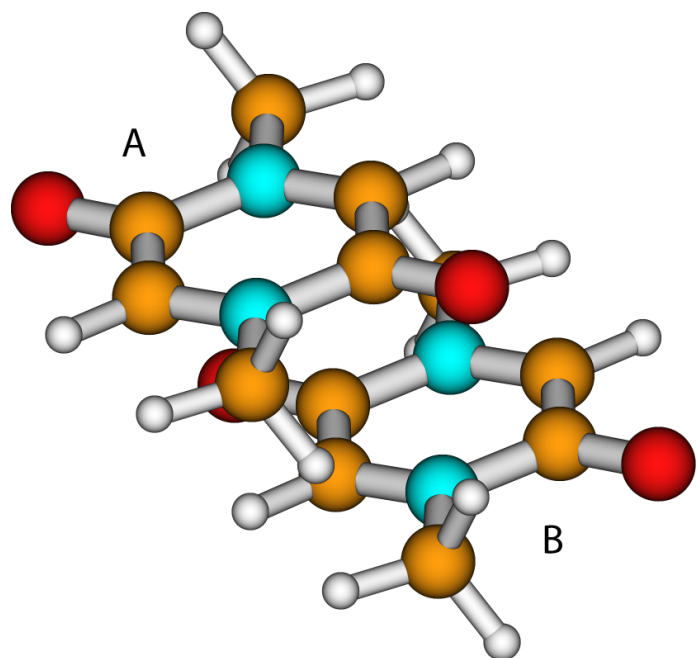
- Proper ASP of molecular wavefunctions
 - Clear chemical interpretation of the state
 - No orthogonalizations



The many-electron basis functions

Löwdin orthogonalise the orbitals
and perform CASCI(12,10):

E (eV)



Ψ_A^0	×	Ψ_B^0	0.00
Ψ_A^1	×	Ψ_B^0	2.58
Ψ_A^0	×	Ψ_B^1	2.58
Ψ_A^T	×	Ψ_B^T	1.44
Ψ_A^+	×	Ψ_B^-	3.56
Ψ_A^-	×	Ψ_B^+	3.56



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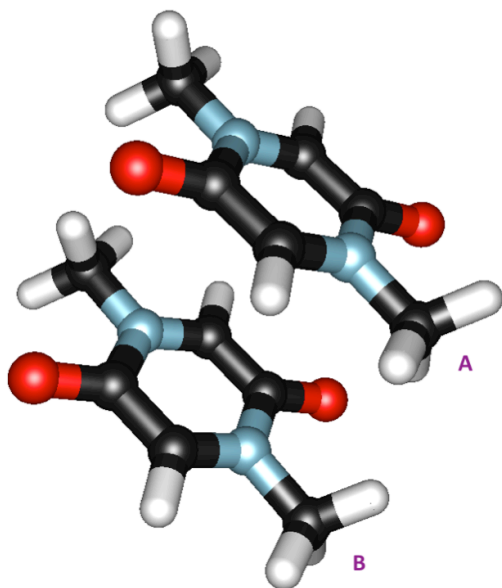
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Theoretical Chemistry

Stack A – GNOME procedure (CAS(2,2))

- The relative energies of the cluster MEBFs



N-N = 3.579 Å

N-N = 3.854 Å

MEBF	$E_{\text{rel}}(\text{eV})$	MEBF	$E_{\text{rel}}(\text{eV})$
$\hat{A}[\text{AB}]$	0.00	$\hat{A}[\text{A}^{\text{T}}\text{B}^{\text{T}}]$	1.61
$\hat{A}[\text{A}^{\text{S}}\text{B}]$	3.81	$\hat{A}[\text{A}^{\text{+}}\text{B}^{\text{-}}]$	4.49
$\hat{A}[\text{AB}^{\text{S}}]$	3.81	$\hat{A}[\text{A}^{\text{-}}\text{B}^{\text{+}}]$	4.49

- The energies and weights of the diabatic states

MEBF	S[1]	S[2]	S'[1]	S'[2]
$\hat{A}[\text{A}^{\text{S}}\text{B}]$	0.50	0.50	0.45	0.50
$\hat{A}[\text{AB}^{\text{S}}]$	0.50	0.50	0.45	0.50
$\hat{A}[\text{A}^{\text{+}}\text{B}^{\text{-}}] + \hat{A}[\text{A}^{\text{-}}\text{B}^{\text{+}}]$	-	-	0.09	0.00
Energy (eV)	3.72	3.90	3.64	3.90

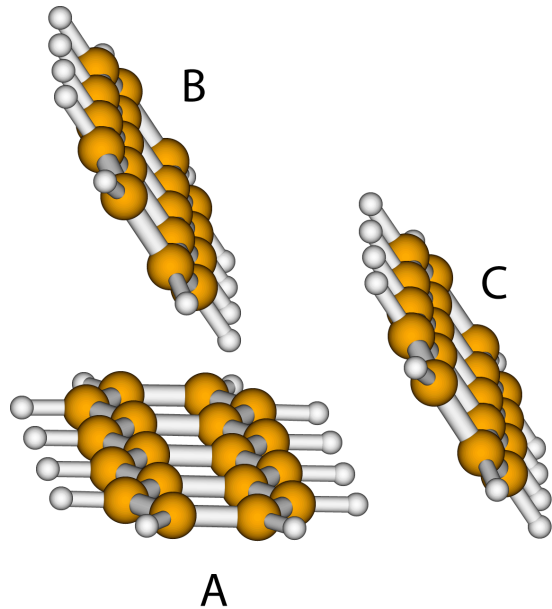


Electronic coupling between diabatic states (meV)

		$\Psi(S[1])$	$\Psi(S[2])$	$\Psi'(S[1])$	$\Psi'(S[2])$
Stack A	$\Psi(^1TT)$	356.7	0.0	353.9	0.1
	$\Psi'(^1TT)$	356.8	0.0	367.6	0.1
Inter-stack	$\Psi(^1TT)$	92.5	23.4	96.6	23.2
	$\Psi'(^1TT)$	92.7	23.9	98.4	23.1
Stack A	$\Psi(^1TT)$	4.0	0.0	16.5	0.0
	$\Psi'(^1TT)$	6.9	0.0	11.8	0.0
Inter-stack	$\Psi(^1TT)$	0.4	0.1	0.2	0.3
	$\Psi'(^1TT)$	3.3	0.0	0.1	0.2



Tetracene trimer



- Coupling between S_1 excited states:
- In dimer:
 - AB: 4.5 meV
 - BC: 43.8 meV
- In trimer:
 - AB: 4.0 meV
 - BC: 44.1 meV
 - AC: 11.6 meV



Financial support

- Zernike Institute for Advanced Materials (Dieptestrategie)
- FOM Focus Group 'Next Generation Organic Photovoltaics' (DIFFER)
- FOM/Shell
- ITN-EJD-TCCM (Horizon2020)



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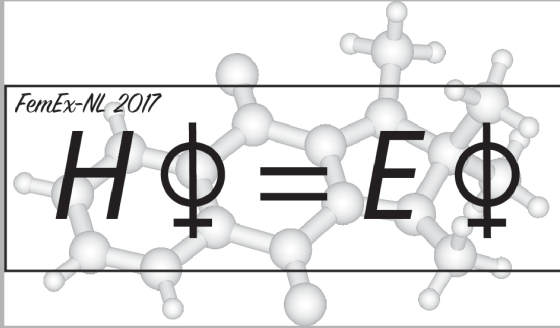
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Theoretical Chemistry


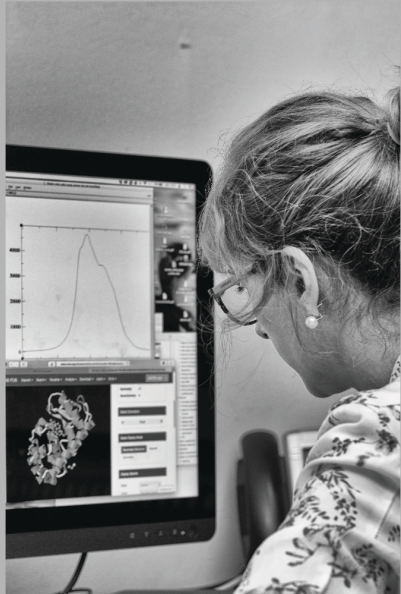
FemEx-NL June 22-25 2017



FemEx-NL 2017

$$H\Phi = E\Phi$$

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Promoting female excellence
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Organization

Celia Fonseca Guerra
Paola Gori-Giorgi
Jocelyne Vreede
Ria Broer
Remco Havenith

Venue:

The Netherlands
Near Amsterdam

<http://theochem.chem.rug.nl/femex/femex2017.html>



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Theoretical Chemistry



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Theoretical Chemistry

$$I_1 = \langle \Delta_a | \sum_{\mu} \Omega_{\mu} | \Delta_b \rangle$$

$$I_1 = \sum_i \sum_j \langle a_i | \Omega_1 | b_j \rangle S(i, j)$$

$$I_2 = \langle \Delta_a | \sum_{\nu > \mu} \Omega_{\mu\nu} | \Delta_b \rangle$$

$$I_2 = \sum_{k > i} \sum_{l > j} \langle a_i a_k | \bar{\Omega}_{12} | b_j b_l \rangle S(ik, jl)$$

For non-singular overlap matrices S:

$$S(ik, jl) = 2(1 - p_{ik}) S(ij) S(kl) / |\mathbf{S}|$$

For singular S a similar factorization is possible, so:

$$I_2 = \sum_{p,q,r,s} \langle \chi_p \chi_r | \Omega_{12} | \chi_q \chi_s \rangle B(pr, qs)$$

where also B can be factorized.

