



# **Multicenter Bond Index: a versatile tool to characterize electron delocalization and aromaticity.**

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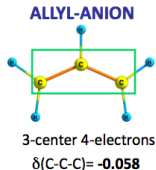
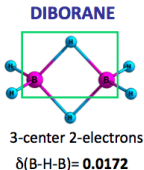
## Acknowledgements

# Multicenter Bonding

Accounts for electron-sharing between  $n$  centers (atoms).

It is related to the  $n$ -order central moment of the electron population.

$$D(A_1, \dots, A_n) \sim \left\langle \left( \hat{N} - \overline{N} \right)^n \right\rangle_{A_1, \dots, A_n}$$



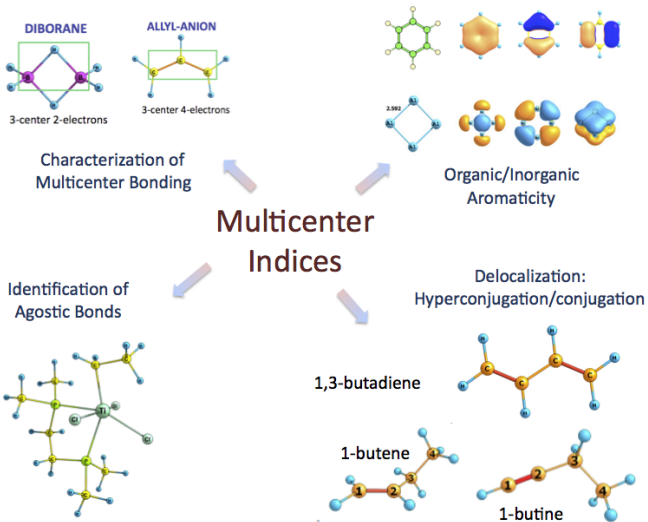
Giambiagi, de Giambiagi, Mundim, *Struc. Chem.* 1, 423 (1990)

Ponec, Mayer, *JPCA* 101, 1738 (1997)

Bochicchio, Ponec, Torre, Lain, *TCA* 105, 292 (2001)

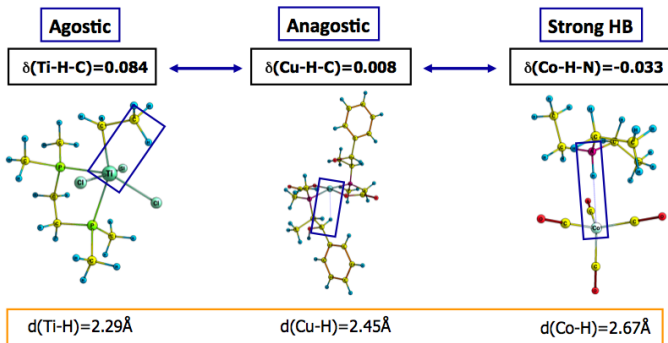
nc-ESI: What is it good for?

# nc-ESI: What is it good for?



## Agostic bonds

# Agostic Bonds



Interaction	3-center ESI
Agostic Bond (3c-2e)	3c-ESI > 0
Anagostic Interaction	3c-ESI $\approx$ 0
Hydrogen Bond (3c-4e)	3c-ESI < 0

Feixas, Matito, Maseras, Poater, Solà, in preparation

## Agostic bonds

# Conjugation and Hyperconjugation effects

Conjugation	butane		1,3-butadiene		1,3-butadiyne	
<b>4c-DI</b>	B3LYP	CCSD	B3LYP	CCSD	B3LYP	CCSD
$\delta(C_1-C_2-C_3-C_4)$	-0.0045	-0.0034	-0.0798	-0.0452	-0.1678	-0.0929
$\delta_\pi(C_1-C_2-C_3-C_4)$			-0.0775		-0.1675	

Hyperconjugation	butane		1-butene		1-butine	
<b>4c-DI</b>	B3LYP	CCSD	B3LYP	CCSD	B3LYP	CCSD
$\delta(C_1-C_2-C_3-C_4)$	-0.0045	-0.0034	-0.0133	-0.0085	-0.0182	-0.0114
$\delta(C_1-C_2-C_3-H_5)$	-0.0005	-0.0003	-0.0013	-0.0009	-0.0191	-0.0121
$\delta(C_1-C_2-C_3-H_6)$	-0.0005	-0.0003	-0.0142	-0.0092	-0.0191	-0.0121

Feixas, Matito, Poater, Solà, JPCA 115, 13104 (2011)

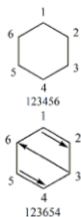
# Aromaticity

Let  $\mathcal{A} = \{A_1, A_2, \dots, A_n\}$  represent a ring.

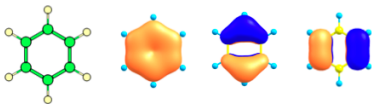
$$I_{ring}(\mathcal{A}) = \sum_{i_1, i_2, \dots, i_n} S_{i_1 i_2}(A_1) S_{i_2 i_3}(A_2) \dots S_{i_n i_1}(A_n)$$

$$MCI(\mathcal{A}) = \sum_{\mathcal{P}(\mathcal{A})} I_{ring}(\mathcal{A})$$

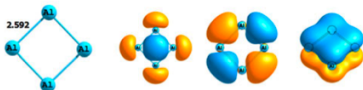
The  $n^{th}$  root of these quantities correlates with the TREPE.



**C<sub>6</sub>H<sub>6</sub>**  
**MCI = 0.071**  
**MCI<sub>σ</sub> = 0.000**  
**MCI<sub>π</sub> = 0.071**

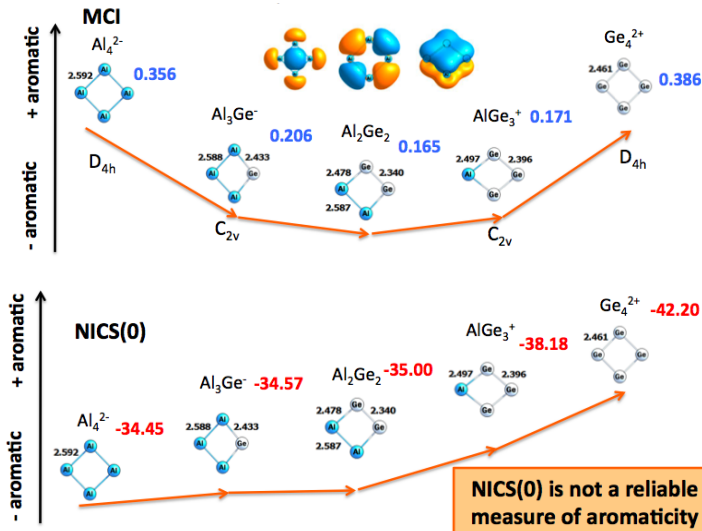


**Al<sub>4</sub><sup>2-</sup>**  
**MCI = 0.356**  
**MCI<sub>σ</sub> = 0.169**  
**MCI<sub>π</sub> = 0.187**



Giambiagi, de Giambiagi, Figuerido, PCCP 2, 3381 (2000)

# Aromaticity

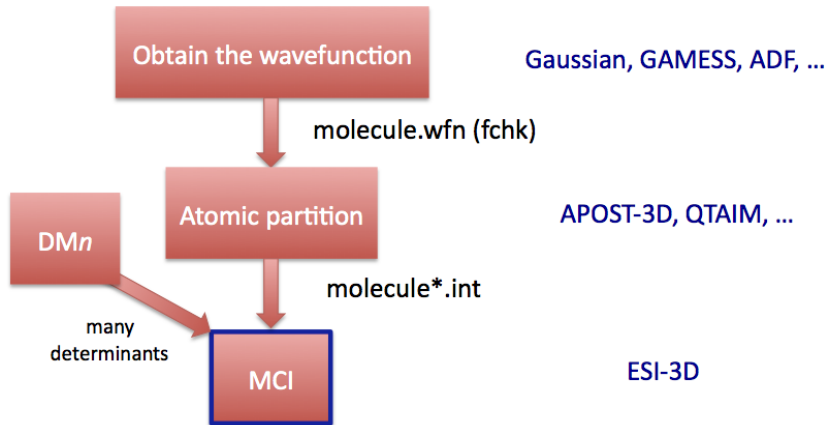






How to obtain nc-ESI?

# Computational Details



## Multicenter Indices: Formulae

Functions that upon integration give the ESIs [assume  $1 \equiv (\vec{r}, s)$ ]:

$$\delta(A, B) = -2 \int_A \int_B d1d2 \gamma(1, 2) = \text{Cov}(N(A), N(B))$$

$$\delta(A, B, C) = 2 \int_A \int_B \int_C d1d2d3 \gamma(1, 2, 3)$$

$$\delta(A_1, A_2, \dots, A_n) = \frac{(-2)^{n-1}}{(n-1)!} \int_{A_1} \int_{A_2} \dots \int_{A_n} d1d2\dots dn \gamma(1, 2, \dots, n)$$

which are

$$\gamma(1, 2) = \rho^{\text{xc}}(1, 2) = \rho(1, 2) - \rho(1)\rho(2)$$

$$\gamma(1, 2, 3) = \rho(1, 2, 3) - \rho(1)\rho(2)\rho(3) - \hat{\mathcal{P}}_{1,2,3}(\rho^{\text{xc}}(1, 2)\rho(3))$$

$$\gamma(1, 2, \dots, n) = \langle (\hat{\rho}_1 - \bar{\rho}_1)(\hat{\rho}_2 - \bar{\rho}_2) \dots (\hat{\rho}_n - \bar{\rho}_n) \rangle$$

they depend on the  **$n$ -order reduced density matrix ( $n$ -RDM)**.

## Multicenter Indices: Bottleneck

The function that gives the  $nc$ -ESI is *symmetric* to the exchange  $1, 2, \dots$

$$\gamma(1, 2, \dots, n) = \langle (\hat{\rho}_1 - \bar{\rho}_1)(\hat{\rho}_2 - \bar{\rho}_2) \cdots (\hat{\rho}_n - \bar{\rho}_n) \rangle$$

and so it is the  $nc$ -ESI calculated. Thus, we cannot calculate  $I_{ring}$ -like expressions with this formulation; only MCI.

Unfortunately,

$$\gamma(1, 2, \dots, n) = \gamma(1, 2, \dots, n)[\rho(1, 2, \dots, n)]$$

*i.e.*, the cost of these indices grows with the size of the  $n$ -RDMs ( $M^{2n}$ ).

**Shall we approximate them?**

## Approximation for the 3-RDM

The HF expression for the three-body density (3-RDM):

$$\rho^{\text{HF}}(1, 2, 3) = \begin{vmatrix} \rho(1) & \rho(1|2) & \rho(1|3) \\ \rho(2|1) & \rho(2) & \rho(2|3) \\ \rho(3|1) & \rho(3|2) & \rho(3) \end{vmatrix}$$

leads to a very simple expression

$$\tilde{\gamma}^{\text{HF}}(1, 2, 3) = 2 \sum_{ijk} \eta_i \eta_j \eta_k \phi_j^*(1) \phi_i^*(1) \phi_k(2) \phi_j^*(2) \phi_i(3) \phi_k^*(3) \rightarrow \tilde{\delta}(A, B, C)$$

In the spirit of Müller approach we suggest:

$$\tilde{\gamma}(1, 2, 3) = 2 \sum_{ijk} (\eta_i \eta_j \eta_k)^{1/2} \phi_j^*(1) \phi_i^*(1) \phi_k(2) \phi_j^*(2) \phi_i(3) \phi_k^*(3) \rightarrow \tilde{\delta}(A, B, C)$$

and

$$\tilde{\gamma}(1, 2, 3) = 2 \sum_{ijk} (\eta_i \eta_j \eta_k)^{1/3} \phi_j(1) \phi_i^*(1) \phi_k(2) \phi_j^*(2) \phi_i(3) \phi_k^*(3) \rightarrow \tilde{\delta}(A, B, C)$$

## Approximations for the 3c-ESI

In the spirit of Müller approach, let us take the simplest natural orbital formulation that integrates to the number of electrons:

$$\tilde{\delta}(A_1, A_2, A_3) = 2 \sum_{i_1, i_2, i_3} n_1^{1/3} n_2^{1/3} n_3^{1/3} S_{i_1 i_2}(A_1) S_{i_2 i_3}(A_2) S_{i_3 i_1}(A_3)$$

We will also include the previous ones:

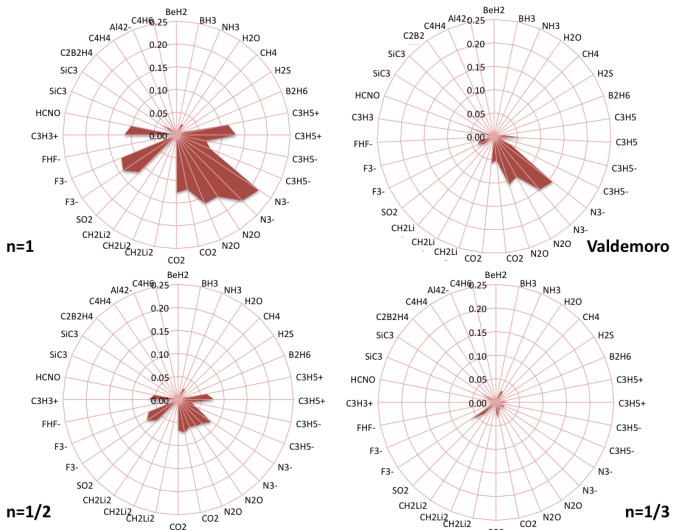
$$\tilde{\delta}(A_1, A_2, A_3) = 2 \sum_{i_1, i_2, i_3} n_1 n_2 n_3 S_{i_1 i_2}(A_1) S_{i_2 i_3}(A_2) S_{i_3 i_1}(A_3)$$

$$\tilde{\delta}(A_1, A_2, A_3) = 2 \sum_{i_1, i_2, i_3} n_1^{1/2} n_2^{1/2} n_3^{1/2} S_{i_1 i_2}(A_1) S_{i_2 i_3}(A_2) S_{i_3 i_1}(A_3)$$

$${}^3D^{Val} = {}^1D^3 + 3({}^2D - {}^1D^2) \wedge {}^1D = 3{}^2D \wedge {}^1D - 2{}^1D^3$$

Similar idea was used in six-member rings, that needs the prohibitive 6-RDM:

# Performance of DM3 approximations



## We suggest a new formula for the 3-RDM:

$$\tilde{\rho}(1, 2, 3) = \tilde{\gamma}(1, 2, 3) + \rho(1)\rho(2)\rho(3) + \hat{\mathcal{P}}_{1,2,3}(\rho^{\text{xc}}(1, 2)\rho(3))$$

where

$$\tilde{\gamma}(1, 2, 3) = \sum_{ijk} \eta_i^{1/3} \eta_j^{1/3} \eta_k^{1/3} \phi_i(1)\phi_j(1)\phi_k(2)\phi_j(2)\phi_i(3)\phi_k(3)$$

- ▶ Fulfills the sum rule.
- ▶ Contains the exact 2-RDM but...
- ▶ ...it does not bring the 2-RDM upon integration of one coordinate.
- ▶ The three-body part is approximated as an expression of NOs.

## Conclusions and prospective work

- ▶ Multicenter indices are connected to several important chemical nomenclatures such as agostic bonds, conjugation and aromaticity.
- ▶ The interpretation is clear but expression depends on huge n-RDMs.
- ▶ Noone goes beyond 3-RDM, and  $n = 1$  (HF-like) approx. is poor.
- ▶ Two new approxs. based on NOs ( $n = 1/2, n = 1/3$ ) have been put forward, these and Valdemoro's (uses 2-RDM!) perform much better than  $n = 1$  (HF-like).



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