

Benoît BRAÏDA

The Maximum Probability Domains method

*Workshop on Topological Approaches to Intermolecular Interactions,
26-28 June 2013, Paris*

Laboratoire de Chimie Théorique

Université Pierre et Marie Curie - Sorbonne universités

Schedule

- 09:00 / 09:30 : The MPD method (BB)
- 16:30 / 17:30 : The MPD program / Training (MM)

Real space regions interpretative methods

<i>Method</i>	<i>Function</i>	<i>Regions</i>
AIM	$\rho(\mathbf{r})$	Atomic regions
ELF	ELF(\mathbf{r})	ELF basins
MPD	$p_v(\Omega)$	Domains of max. probability (Ω)

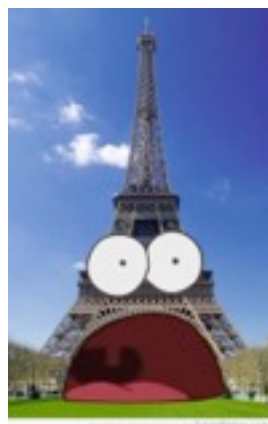
- Ancestor of MPD : Daudel's loges theory

MPD history

- Selected references :
 - Savin A (2002) In: Sen KD (ed) Reviews of modern quantum chemistry: a celebration of the 252 contributions of Robert G. Parr. World Scientific, Singapore, p43
 - Cancès E, Keriven R, Lodier F, Savin A (2004) Theor Chem Acc 111, 373
 - Scemama A, Caffarel M, Savin A (2007) J Comp Chem 28, 442
 - Causa M, Savin A Z. Anorg. Allg. Chem. 2011, 637 / Causa M, Savin A J. Phys. Chem. A 2011, 115, 13139
 - Mafra Lopez Jr. O, Braida B, Causa M, Savin A in Progress in Theoretical Chemistry and Physics vol 22, p173 ed Hoggan, Springer UK, London (2011)

MPD connexion

Paris



Napoli



Mauro Causa'

MPD program (solids)

Andreas Savin, Benoît BRAÏDA
Pascal Frey, Simona B. Savescu
MPD program (molecules)

Ghent



Patrick Bultinck
Guillaume Acke

Oviedo



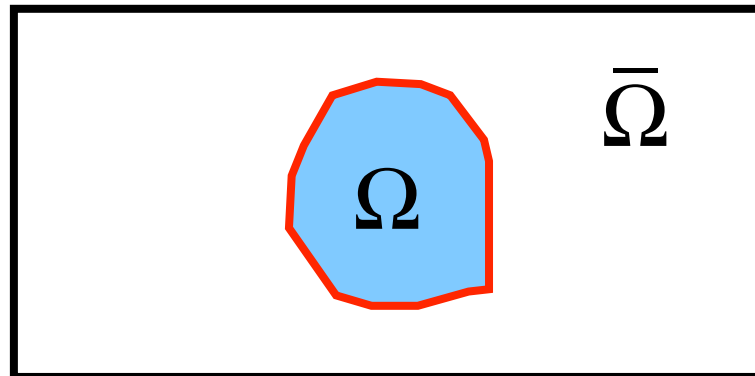
Ángel Martín Pendás
Marcos Menendez

Basics

1) The probability function :

Probability to find ν *and only* ν electrons in a given region of space Ω :

$$p_\nu(\Omega) = \binom{N}{\nu} \int_{\Omega} dx_1 \dots dx_\nu \int_{\bar{\Omega}} dx_{\nu+1} \dots dx_N |\psi|^2$$



Basics

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Rem : $p_1(\Omega) \neq \int_{\Omega} \rho(x_1) dx_1 = N \int_{\Omega} dx_1 \int_{R^3} dx_2 \dots dx_N |\psi|^2 = \langle \nu \rangle_{\Omega}$

We rather have : $\langle \nu \rangle_{\Omega} = \sum_{\nu=0}^N \nu p_\nu(\Omega)$

Basics

1) The probability function :

Example : dihydrogen molecule - large interatomic distance :

	$p_1(\Omega)$	$p_0(\Omega) = p_2(\Omega)$	$\langle v \rangle$
<i>Ground (covalent) state :</i> $\text{H} \cdot \text{---} \cdot \text{H}$	1.	0.	1.
<i>Excited ionic state :</i> $\left\{ \begin{array}{l} \ominus \text{H} : \\ \oplus \text{H} \end{array} \right. \left. \begin{array}{l} : \text{H} \oplus \\ \ominus \text{H} \end{array} \right.$	0.	0.5	1.

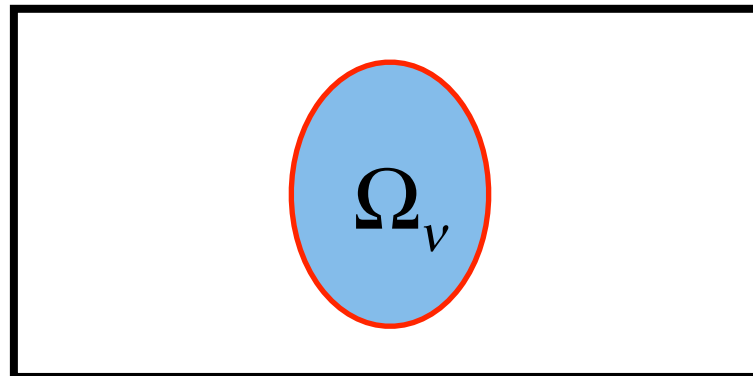
➡ Probabilities contain more informations than populations

Basics

2) MPD / Definition :

A **Maximum Probability Domain (MPD)** is a region of space *locally* maximizing $p_v(\Omega)$:

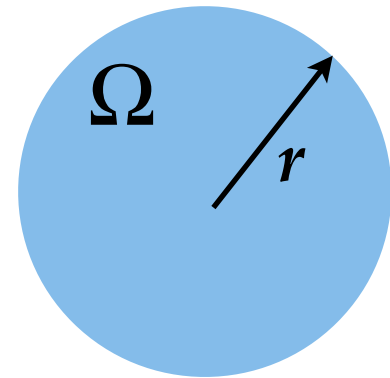
$$\Omega_v = \max_{\Omega} p_v(\Omega)$$



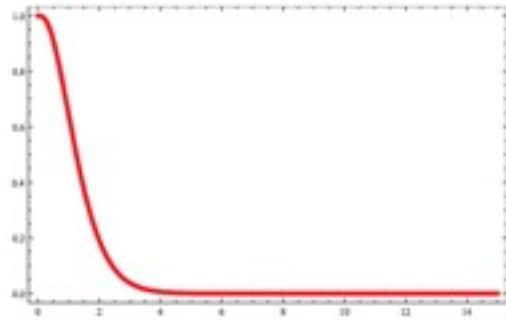
Basics

2) MPD / existence :

Example : atom case ($N e^-$),
atomic-centered spherical domain :



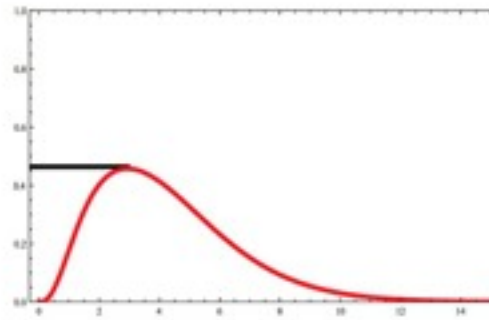
p_0



$\nu = 0$

r

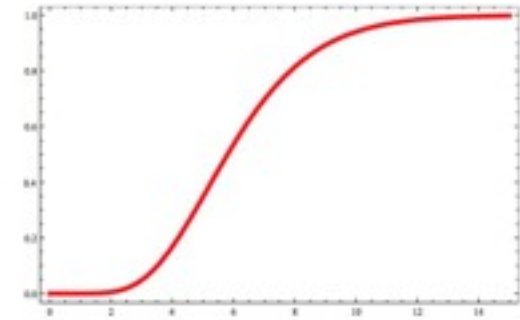
p_ν



$0 < \nu < N$

r

p_N



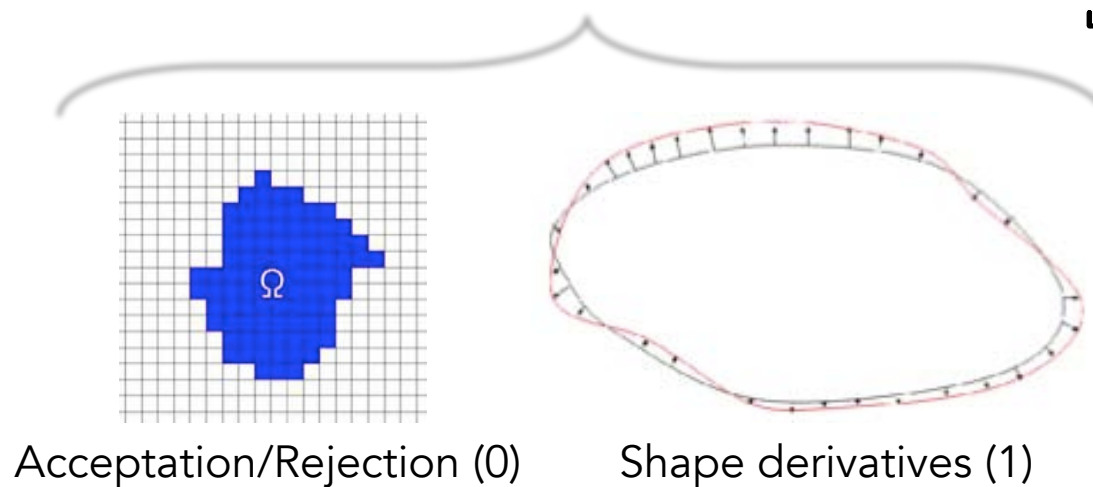
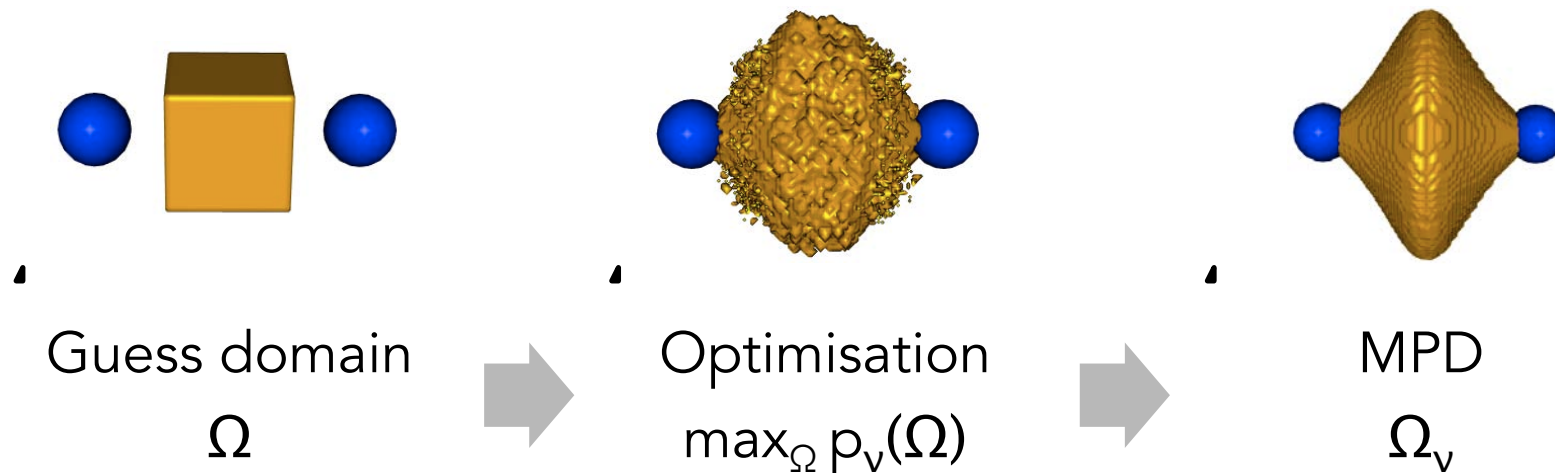
$\nu = N$

r

➡ For any ν , at least one Ω_ν always exists

Basics

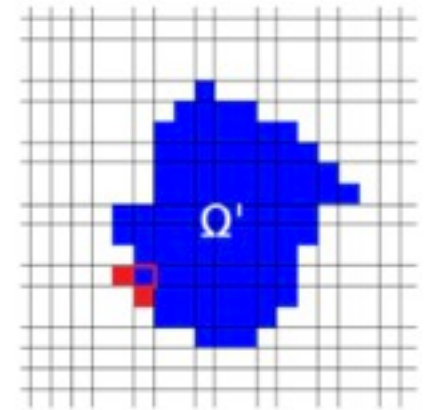
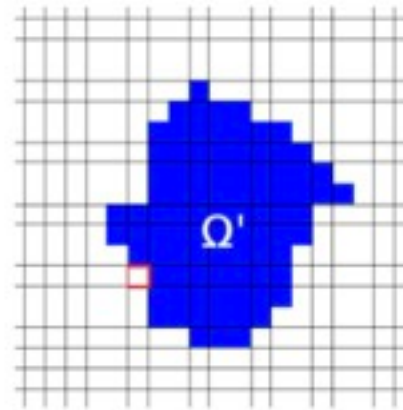
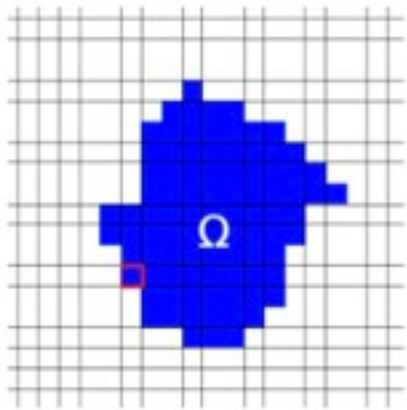
2) MPD / Optimization (in a nutshell) :



Basics

2) MPD / Optimization (in a nutshell) :

Acceptation/rejection algorithm (0) :



Random element selection of the surface

$$\rho_\nu(\Omega') > \rho_\nu(\Omega) \implies \text{OK}$$

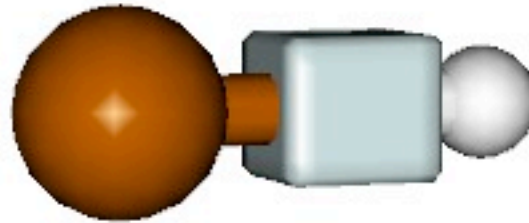
$$\rho_\nu(\Omega') > \rho_\nu(\Omega) \implies \text{OK}$$

→ Single domain optimization only

Basics

2) MPD / Optimization (example) :

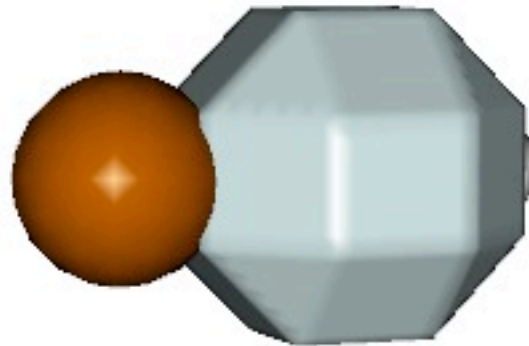
Searching a Ω_2 domain in the Li—H molecule :



Basics

2) MPD / Optimization (example) :

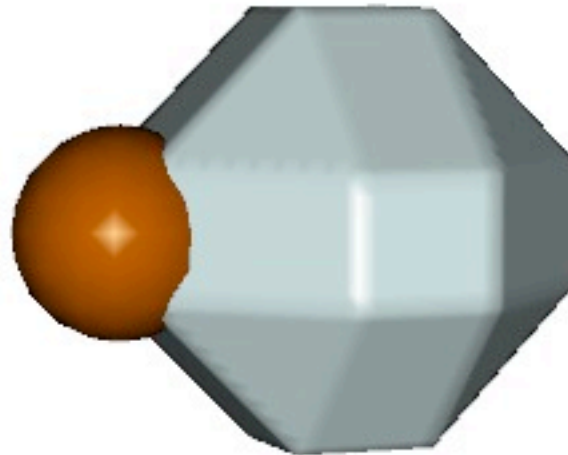
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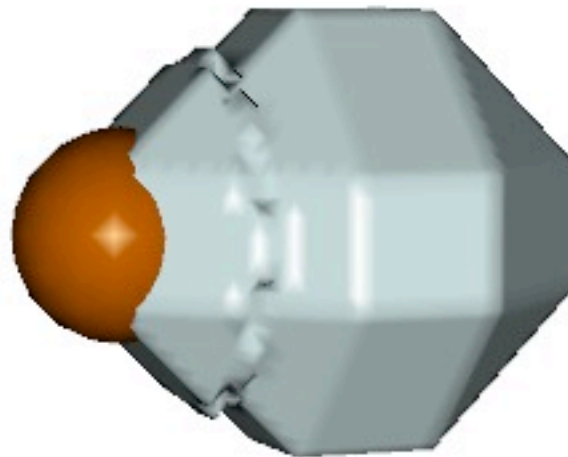
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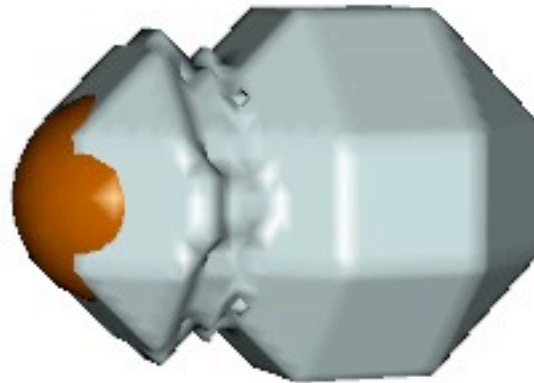
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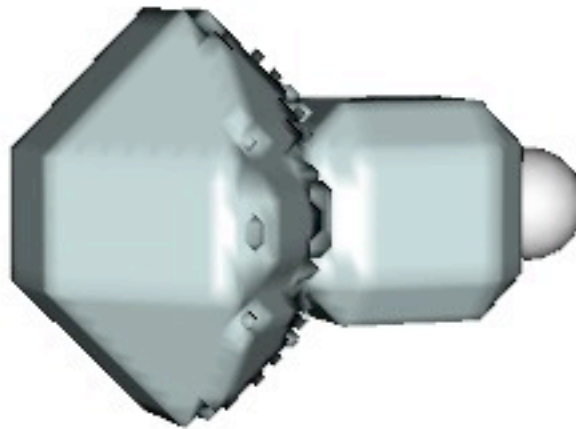
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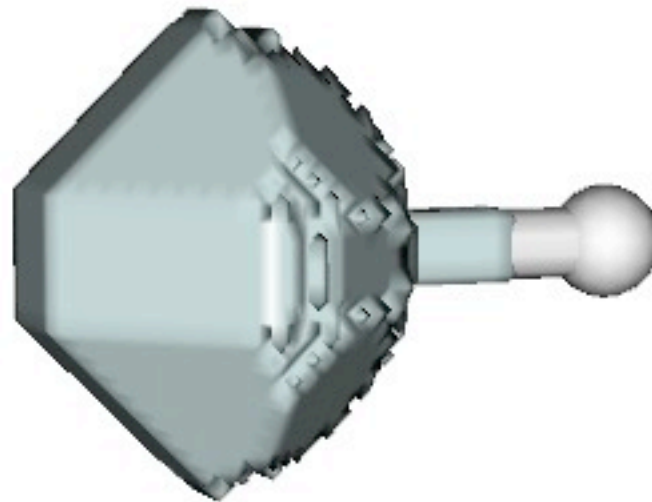
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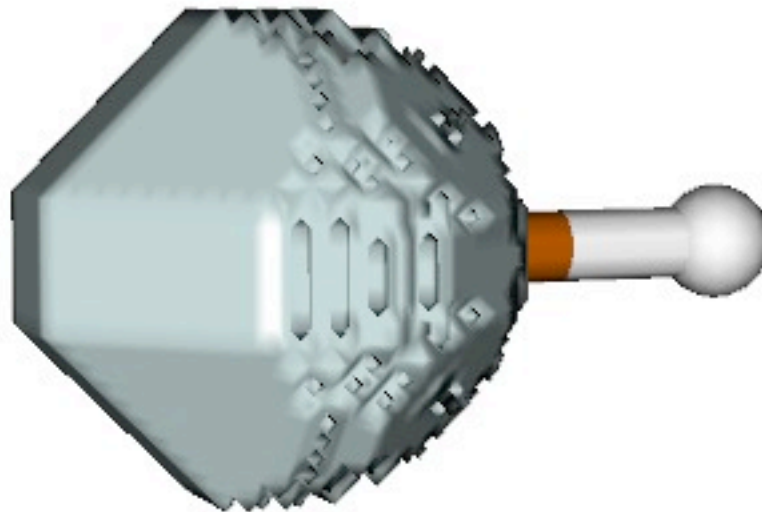
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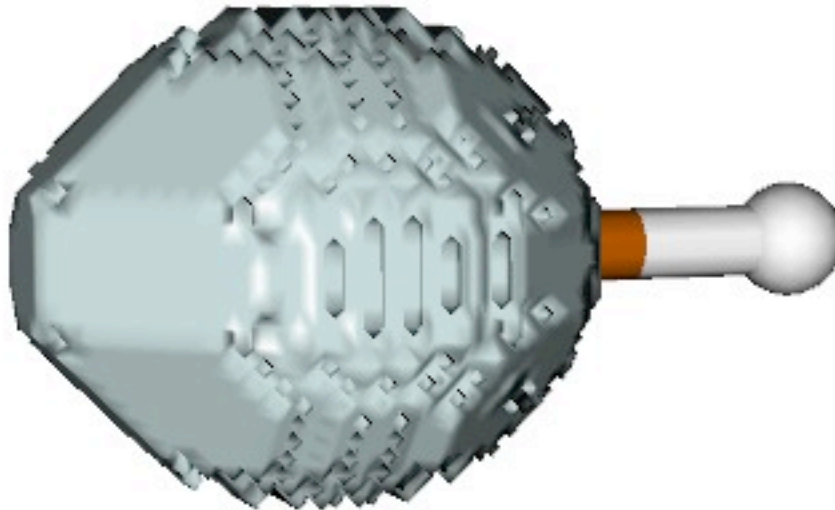
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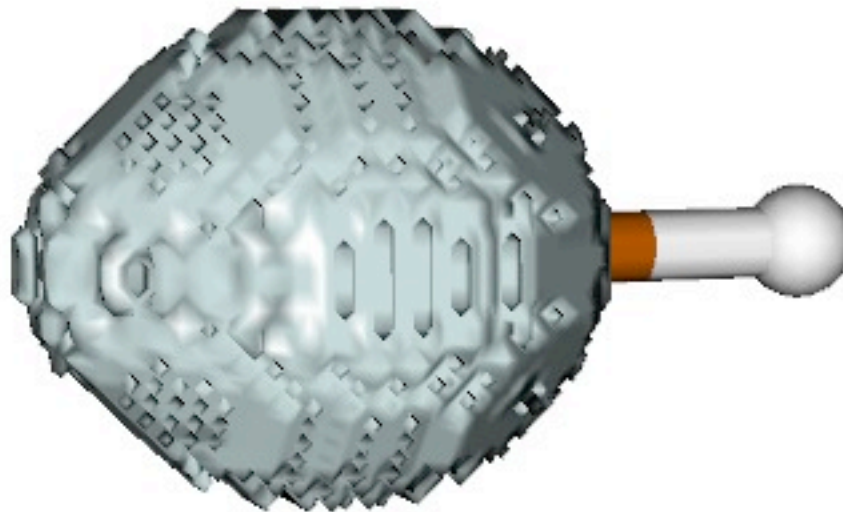
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Basics

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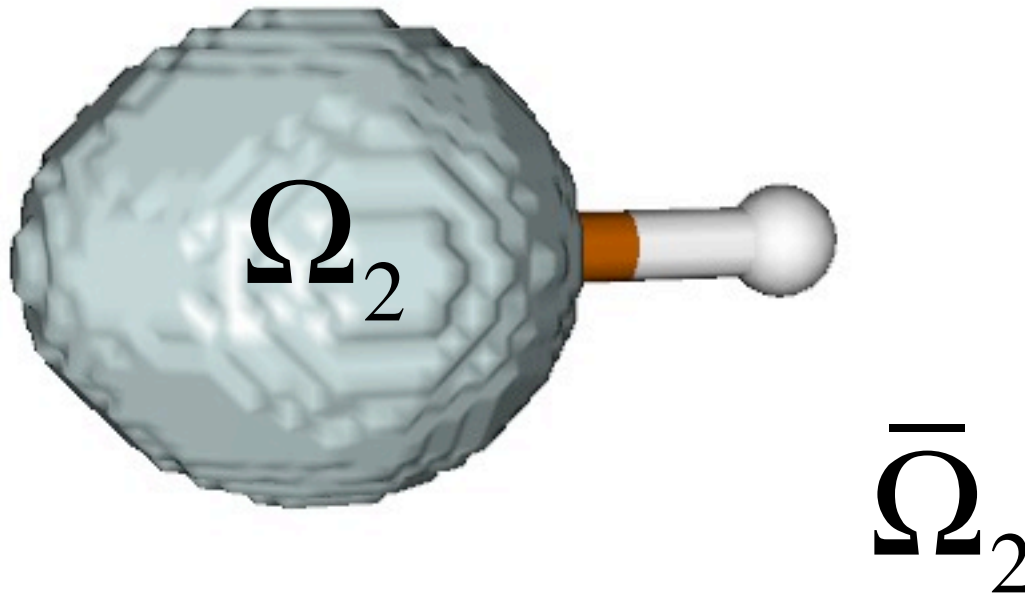
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Basics

2) MPD / Optimization (example) :

Searching a Ω_2 domain in the Li—H molecule :



Examples and properties

- Ne atom (10e⁻) :

$$p_2(\Omega) = \binom{10}{2} \int_{\Omega} dx_1 dx_2 \int_{\bar{\Omega}} dx_3 \dots dx_{10} |\psi|^2 = \binom{8}{2} \int_{\Omega} dx_9 dx_{10} \int_{\bar{\Omega}} dx_1 \dots dx_8 |\psi|^2 = p_8(\bar{\Omega})$$

$\Omega 8$ (valence)



$\Omega 2$ (core)

→ MPDs always provide a partition of space in two parts

Examples and properties

- Ne atom (10e⁻) :

$$p_2(\Omega) = \binom{10}{2} \int_{\Omega} dx_1 dx_2 \int_{\bar{\Omega}} dx_3 \dots dx_{10} |\psi|^2$$

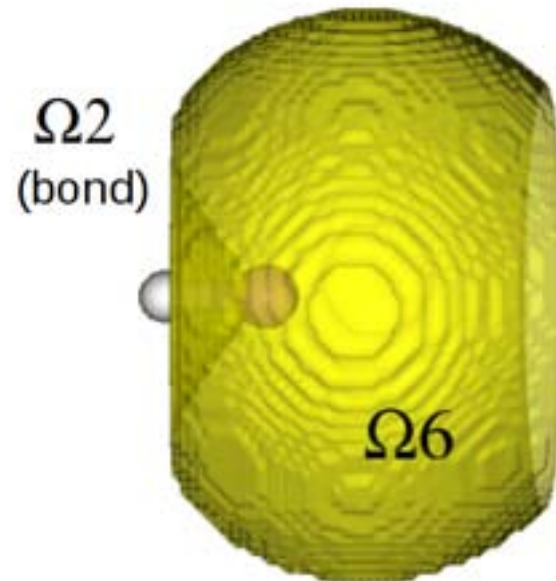


- ➔ Multiple «chemical» solutions may exist (core / valence pairs)
- ➔ Multiple solutions due to symmetry may exist

Examples and properties

- H-F molecule with pseudo ($8e^-$) :

$$p_2(\Omega) = \binom{8}{2} \int_{\Omega} dx_1 dx_2 \int_{\bar{\Omega}} dx_3 \dots dx_{10} |\psi|^2 = p_6(\bar{\Omega})$$

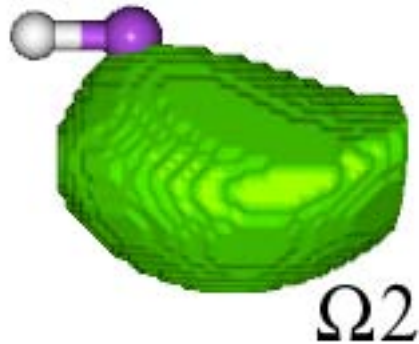


Examples and properties

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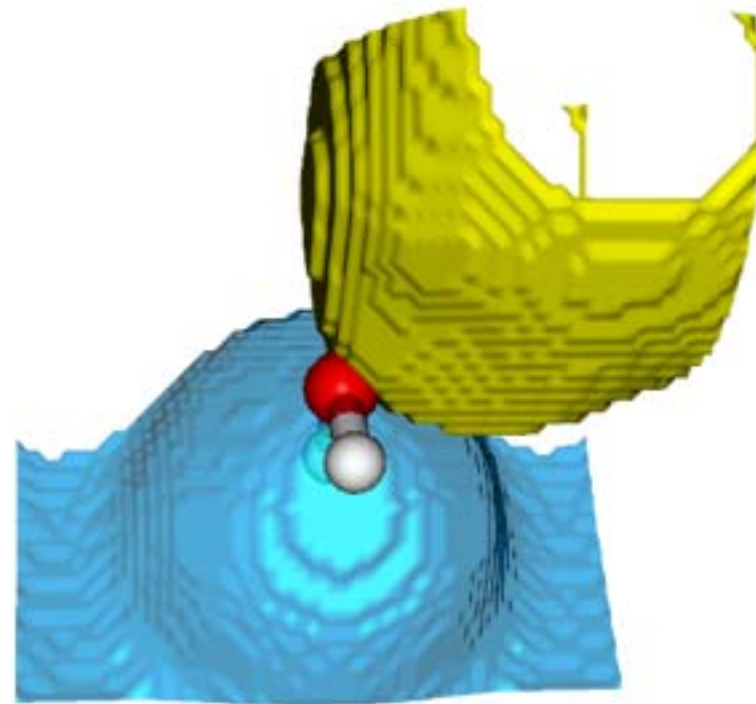
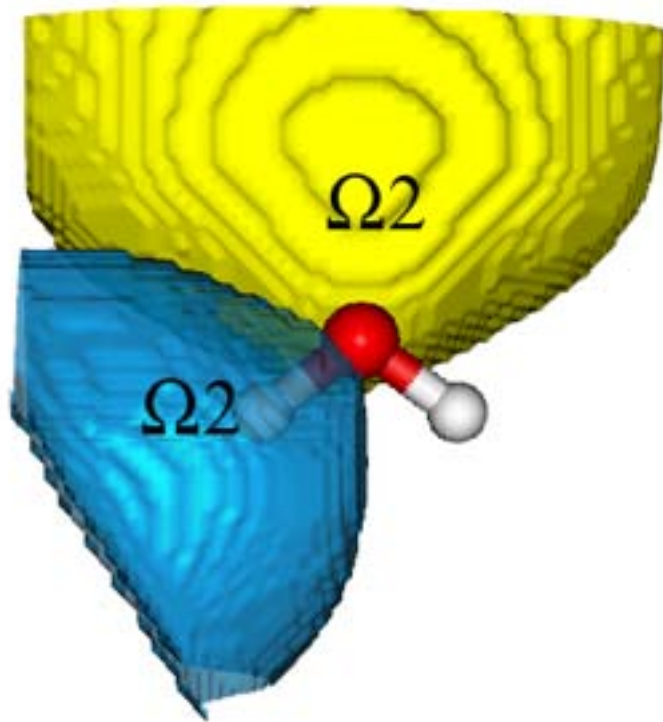
H - F



➔ MPD allows different viewpoints (Ω_v search for any v)

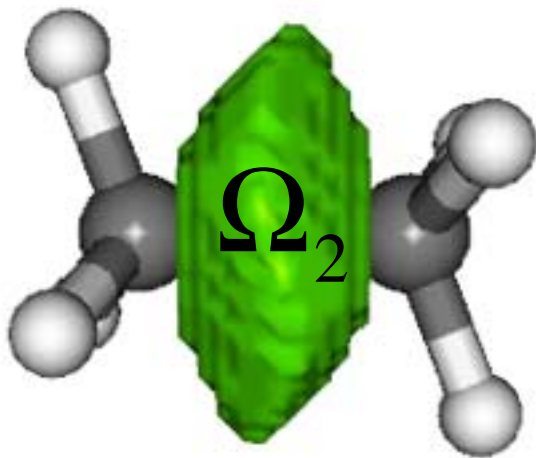
Examples and properties

- Water molecule :



Examples and properties

- Ethane :



$$p_2 = 0.402$$

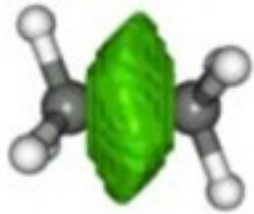
$$\langle \nu \rangle_{\Omega} = 1.965$$

$$\text{Vol.}(\Omega) = 19.050$$

➔ Population is close to ν even if it is p_{ν} which is optimized

Examples and properties

- MPD vs. ELF / Ethane :



$p_2(\Omega)$

$$\begin{aligned} p_2 &= 0.402 \\ \langle \nu \rangle_{\Omega} &= 1.965 \\ \text{Vol.}(\Omega) &= 19.050 \end{aligned}$$



ELF C-C basin

$$\begin{aligned} p_2 &= 0.40 \\ \langle \nu \rangle_{\text{basin}} &= 1.82 \\ \text{Vol.}(\text{basin}) &= 17.66 \end{aligned}$$

➔ ELF basins may be good approximations of MPDs

Examples and properties

- MPD vs. ELF / Dinitrogen :



$p_6(\Omega)$

$$\begin{aligned} p_6 &= 0.307 \\ \langle \nu \rangle_{\Omega} &= 5.989 \\ \text{Vol.}(\Omega) &= 611.842 \end{aligned}$$



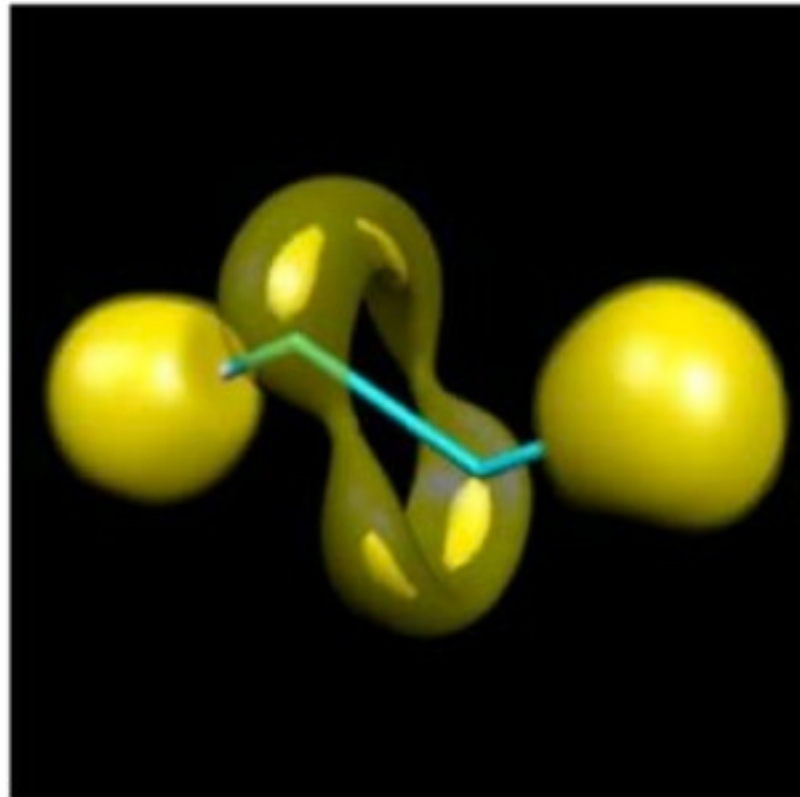
ELF N-N basin

$$\begin{aligned} p_6 &= 0.04 \\ \langle \nu \rangle_{\text{basin}} &= 3.41 \\ \text{Vol.}(\text{basin}) &= 40.66 \end{aligned}$$

→ ELF basins may also be poor approximations of MPDs !

Examples and properties

- MPD vs. ELF / Bent Si_2H_2 :

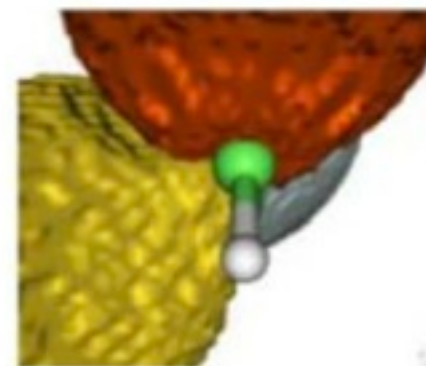
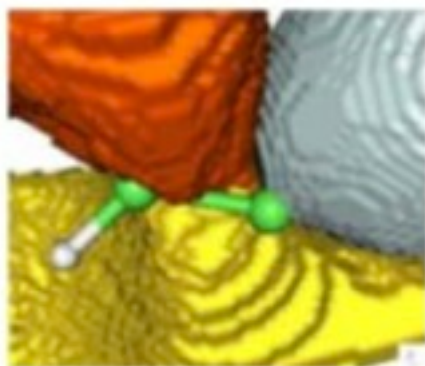
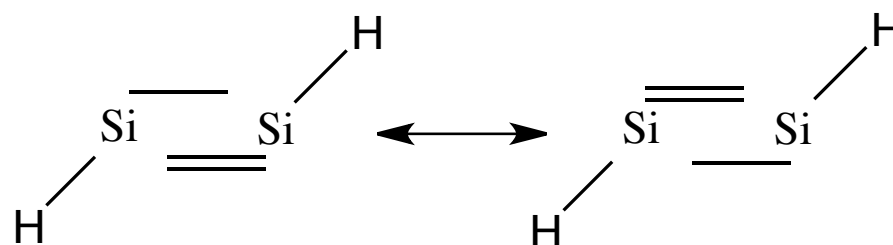


Four ELF
maxima (???)

ELF isosurfaces

Examples and properties

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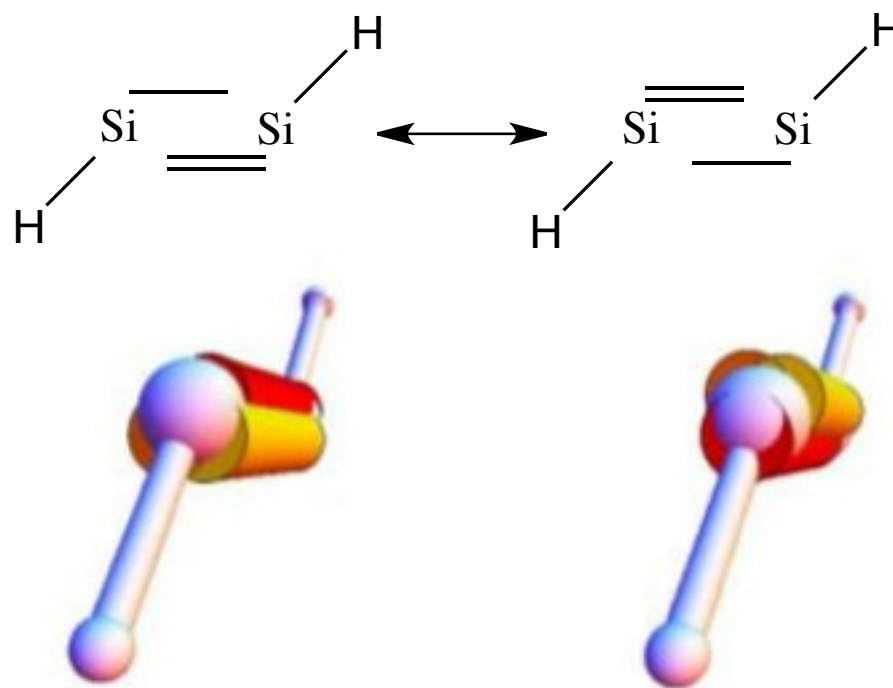


MPDs: Three (shifted) banana bonds

A. Scemama et al., *J. Comp. Chem* 28, 442 (2007)

Examples and properties

- MPD vs. ELF / Bent Si_2H_2 :

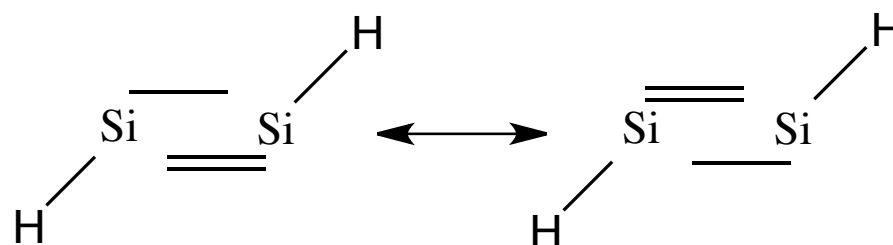


MPDs: Two equivalent triples of banana bonds

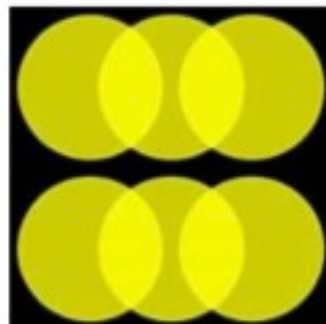
A. Scemama et al., *J. Comp. Chem* 28, 442 (2007)

Examples and properties

- MPD vs. ELF / Bent Si_2H_2 :

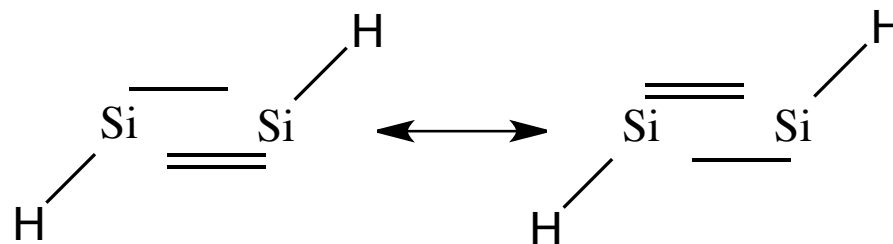


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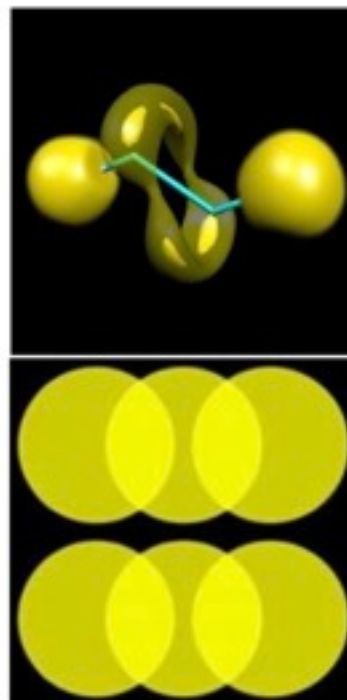


Examples and properties

- MPD vs. ELF / Bent Si_2H_2 :



A. Scemama et al., J. Comp. Chem. 28, 442 (2007)



Perspectives

- **Algorithms developments** (adaptative mesh, level sets,...)
- **Conceptual developments** (compute physical quantities into MPDs, volume-related definition of hardness,...)
- **Applications** (hypervalency,...)