

Fluctuation of electron populations and chemical bonding

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Introduction

What drives chemical bonding?

- $E(A - B) < E(A) + E(B)$
 - ▶ $E = T + V_{ne} + V_{ee}$. Subtle effects: $\Delta E \ll E$.
 - ▶ Solutions? Large contributions to E remain dominated by nuclei \approx constant.
 - ▶ Partitioning E in chemically meaningful ways.
- Orbital paradigm: $1e$ picture.
 - ▶ $\Psi \approx |\phi_1 \dots \phi_N|$. $\phi_i = \sum_{\mu} c_{i\mu} \chi_{\mu}$.
 - ▶ $T = \sum t_i \dots$

Lesson: No bonding without overlap, $\langle \chi_{\nu} | \chi_{\mu} \rangle = S_{\nu\mu} \neq 0$



Introduction

What drives chemical bonding? (In real space)

CEJ 13, 9362, (2007)

- Real space analyses: 3D regions with chemical meaning. $R^3 = \bigcup_A \Omega_A$

$$E = \sum_A \overbrace{\{T_A + V_{ee}^{AA} + V_{en}^{AA}\}}^{E_{self}^A} + \sum_{A>B} \overbrace{\{V_{en}^{AB} + V_{ne}^{AB} + V_{nn}^{AB} + V_{ee}^{AB}\}}^{E_{int}^{AB}}$$

$$V_{ee}^{AB} = \int_{\Omega_A} \int_{\Omega_B} d\mathbf{r}_1 d\mathbf{r}_2 \rho_2(\mathbf{r}_1, \mathbf{r}_2)/r_{12}, \quad \rho_2 = \rho(\mathbf{r}_1)\rho(\mathbf{r}_2) - \rho_{xc}$$

$$E_{int}^{AB} = V_{cl}^{AB} + V_{xc}^{AB}, \quad E = \sum_A E_{self}^A + \sum_{A>B} E_{int}^{AB}$$

Lesson: No bonding without $V_{xc}^{AB} = \int_{\Omega_A} \int_{\Omega_B} d\mathbf{r}_1 d\mathbf{r}_2 \rho_{xc}(\mathbf{r}_1, \mathbf{r}_2)/r_{12}$

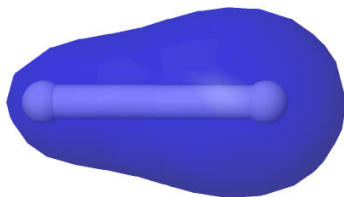
Introduction

What drives chemical bonding? (In real space)

- $\rho_{xc}(1, 2) = \rho(1)(\rho(2) - \rho(2|1)) = \rho(1)\rho_h(2|1)$
- $V_{xc}^{AB} = \int_{\Omega_A} \int_{\Omega_B} d1d2 \rho(1)\rho_h(2|1)/r_{12}$
- The hole has to be delocalized.

No bonding without (hole) delocalization

$\text{H}_2 \rho_h(2|\text{rightH}) r = 1.4 \text{ a.u.}$



Introduction

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No bonding without (hole) delocalization

$\text{H}_2 \rho_h(2|\text{rightH}) r = 6.0 \text{ a.u.}$



Introduction

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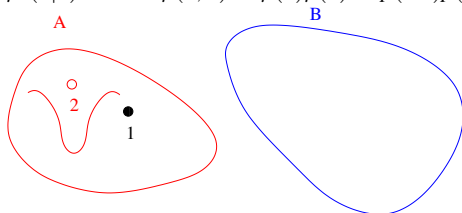
- ρ_{xc} admits a probabilistic interpretation (infinitesimal $d1, d2$).
 - ▶ $\rho(1)d1 \equiv$ P. of finding **one and only one** electron in $d1$.
 - ▶ $\rho_2(1, 2)d1d2 \equiv$ P. **one and only one** electron in $d1$, **one and only one** electron in $d2$
 - ▶ Sample space in $d1, d2$?: $S = \{(0_{d1}, 0_{d2}), (1_{d1}, 0_{d2}), (0_{d1}, 1_{d2}), (1_{d1}, 1_{d2})\}$.
 - ▶ $\rho_{xc}(1, 2)d1d2 = (\rho(1)\rho(2) - \rho_2(1, 2))d1d2 = p(1_{d1})p(1_{d2}) - p(1_{d1}, 1_{d2})$
 - ▶ $\rho_{xc}(1, 2)d1d2 = \sum_S p(S)n_1(S) \sum_S p(S)n_2(S) - \sum_S p(S)n_1(S)n_2(S)$
 - ▶ $\rho_{xc}(1, 2)d1d2 = \langle n_1 \rangle \langle n_2 \rangle - \langle n_1 n_2 \rangle = -\text{cov}(n_1, n_2)$.
 - ▶ ρ_{xc} is a covariance density.
- General: $\int_{\Omega_A} \int_{\Omega_B} d1d2 \rho_{xc}(1, 2) = -\text{cov}(n_A, n_B) = -\langle (n_A - \langle n_A \rangle)(n_B - \langle n_B \rangle) \rangle$
 - ▶ Fluctuation of population: $\text{cov}(n_A, n_B) = \langle \delta_A \delta_B \rangle$
 - ▶ Similarly, $\int_{\Omega_A} \int_{\Omega_A} d1d2 \rho_{xc}(1, 2) = \langle n_A \rangle - \text{var}(n_A)$



Introduction

Delocalization & statistical independence

- Imagine two regions A, B .
- $\rho_h(2|1) \in A \forall 1 \in A$ (hole strictly localized in A)
 - ▶ Knowing an electron lies in A decreases p of finding another only in A .
 - ▶ $\text{cov}(n_A, n_B) = 0, \text{var}(n_A) = 0$.
 - ▶ A and B populations are statistically isolated.
 - ▶ $\rho_h(2|1) = 0 \Rightarrow \rho(1, 2) = \rho(1)\rho(2) \equiv p(1_{d1})p(1_{d2}) = p(1_{d1}, 1_{d2})$



- Despite general **dependence** \Rightarrow blocks of independent e 's.
- Study the distribution of electron population in coarse-grained regions.
- Finding such regions \equiv finding chemically distinctive blocks.



Electron Population Distribution Functions

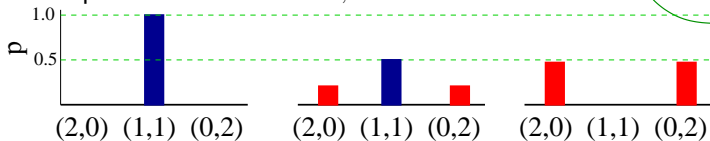
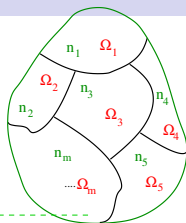
Population Statistics

PCCP 9, 1087 (2007)

- N electrons distributed in m regions. $R^3 = \Omega_1 \cup \Omega_2 \cup \dots \cup \Omega_m$

- $p(n_a, n_b, \dots, n_m) = \frac{N!}{n_a! n_b! \dots n_m!} \int_a d1 \int_a d2 \dots \int_m dN \Psi^*(1, \dots, N) \Psi(1, \dots, N)$

- Two equivalent boxes: $N = 2$, $m = 2$



$$\text{var}(n_a) = \text{cov}(n_a, n_b) = 0 \quad \text{var}(n_a) = -\text{cov}(n_a, n_b) = 0.5 \quad \text{var}(n_a) = -\text{cov}(n_a, n_b) = 1.0$$

- $N_a = \langle n_a \rangle$, $N_{aa} = N_a - \text{var}(n_a)$, $N_{ab} = -\text{cov}(n_a, n_b)$, $\delta^{AB} = 2N_{ab}$

- $H_2?$

- ▶ Simple partition
- ▶ Hartree-Fock exactly binomial (0.25,0.5,0.25)
- ▶ From holes to models: Two independent electrons.
- ▶ Full-CI: (0.21,0.57,0.21).



Electron Population Distribution Functions

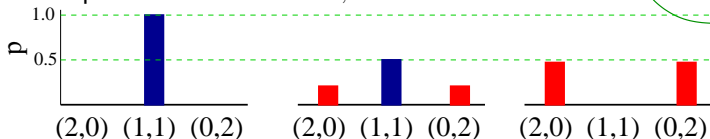
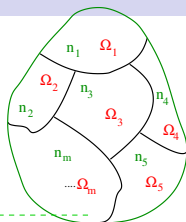
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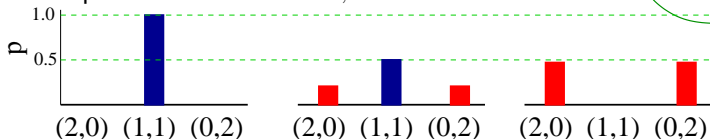
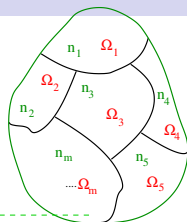
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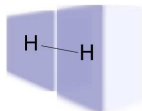


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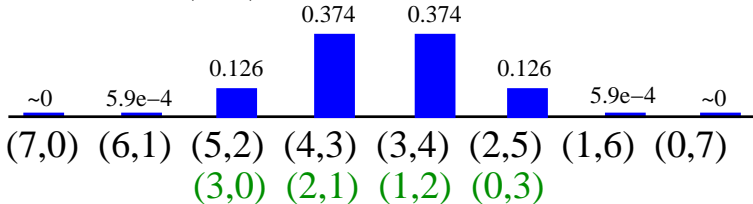
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- $H_2?$

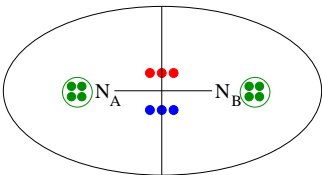
- Simple partition
- Hartree-Fock exactly binomial (0.25, 0.5, 0.25)
- From holes to models: Two independent electrons.
- Full-CI: (0.21, 0.57, 0.21).



- At the HF level, $p(n_a, n_b) = p^\alpha \otimes p^\beta$



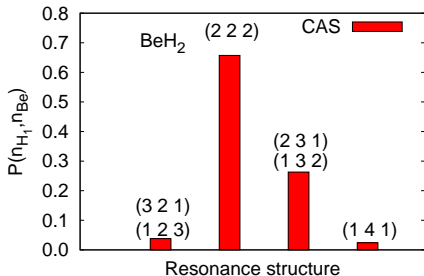
- 3 independent $1e$ p 's ($p_1(0, 1) = p_1(1, 0) = 0.5$) provide: $p_3 = p_1 \otimes p_1 \otimes p_1$
 $p(3, 0) = p(0, 3) = (1/2)^3 = 0.125$, $p(2, 1) = p(1, 2) = 0.375$



- 4 extremely localized electrons in each basin.
- 3 α , 3 β fully delocalized electrons.
- Thus... 3 bonding pairs:
 $\delta = -2\text{cov}(n_a, n_b) = 3.038$
- Correlation (CAS[10,10]) hinders delocalization: $\delta = 1.987$

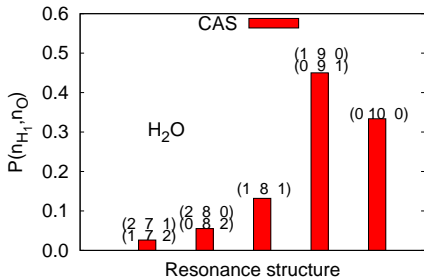


EDF examples: Simple QTAIM full EDFs



BeH₂

- 7:3 H⁻Be²⁺H⁻ : H-Be⁺H⁻ ↔ H⁻Be⁺-H
- (2 3 1) very different from (3 2 1): Chemistry
- Partial statistics available:
 - ▶ $\delta^{Be,H} = 0.302$, but $\delta_0^{Be,H} = 0.522$
- in diatomic BeH, $\delta^{Be,H} = 0.460$

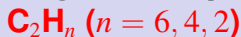


H₂O

- Less ionic than BeH₂
- 1e transfers dominate.. $q(O) = -1.087$
- but the O²⁻ contribution is important.
- Partial statistics
 - ▶ $\delta^{O,H} = 0.637$, and $\delta_0^{O,H} = 0.640$



Lego example



- Examine bonding between the CH_3 , CH_2 or CH groups.
- Hypothesis: non-polar single, double, triple bonds.
- The **modeled** α or β EDFs are easily constructed:

$$\text{C}_2\text{H}_6: \left(\frac{1}{2}, \frac{1}{2}\right) = (0.5, 0.5)$$

$$\text{C}_2\text{H}_4: \left(\frac{1}{2}, \frac{1}{2}\right) \times \left(\frac{1}{2}, \frac{1}{2}\right) = (0.25, 0.50, 0.25)$$

$$\text{C}_2\text{H}_2: \left(\frac{1}{2}, \frac{1}{2}\right) \times \left(\frac{1}{2}, \frac{1}{2}\right) \times \left(\frac{1}{2}, \frac{1}{2}\right) = (0.125, 0.375, 0.375, 0.125)$$

- ... to be compared with the **computed** HF data.

$$\text{C}_2\text{H}_6: (\dots, 0.0187, 0.4811, 0.4811, 0.0187, \dots)$$

$$\text{C}_2\text{H}_4: (\dots, 0.061, 0.250, 0.4878, 0.250, 0.061, \dots)$$

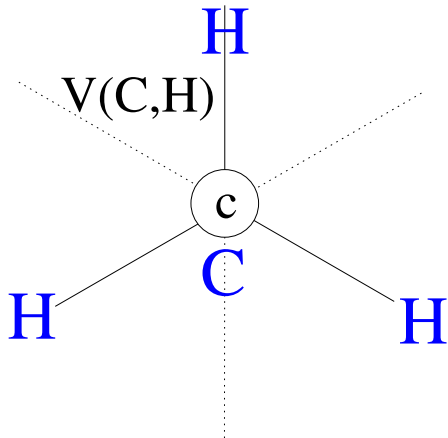
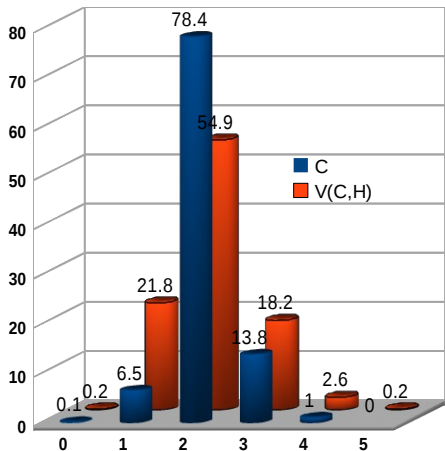
$$\text{C}_2\text{H}_2: (\dots, 0.1251, 0.3747, 0.3747, 0.1251, \dots)$$

- Atomic (or group) regions **seem to isolate quite well groups of localized and delocalized electrons in agreement with chemical wisdom**



EDFs for other space partitions: ELF domains

● CH₄: $p(n_a)$



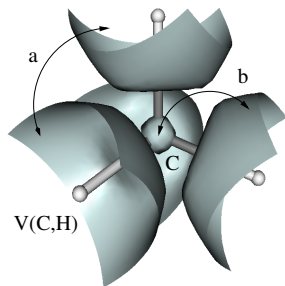
How large is delocalization among localized domains?

- $\delta(CV, VV) = 0.12, 0.37$.
- QTAIM $\delta(CH, HH) = 0.98, 0.04$. Where is simplicity hidden?

p	Core	V(C,H1)	V(C,H2)	V(C,H3)	V(C,H4)
0.2069	2	2	2	2	2
0.0347	2	1	2	2	3
0.0161	3	1	2	2	2
0.0079	2	3	3	1	1
0.0065	1	3	2	2	2
0.0044	2	1	1	2	4
0.0039	3	3	1	2	1
0.0024	2	3	3	2	0
0.0016	1	3	3	2	1
0.0012	3	3	2	2	0

0.94

- Build a model: α set, forget the core
 - ▶ 4 α electrons in the 4 V(C,H) regions.
 - ▶ Elementary events: inter V jumps.
 - ▶ Each electron has a probability j of jumping to any of the other V's
 - ▶ Each electron has a probability s of staying in its own V.
 - ▶ $s + 3j = 1$.



How large is delocalization among localized domains?

JCP 131, 124125 (2009)

- α probs. without core (HF).

p	V1	V2	V3	V4
0.4960	1	1	1	1
0.0377	2	0	1	1
0.0041	2	2	0	0
0.0021	3	1	0	0
0.0000	4	0	0	0

- $s = 0.832$, $j = 0.056$. (Only one parameter).
- Very good fit. **Lessons:**
 - ▶ The small $p(2, 2, 2, 2, 2) \approx 0.21$ value is due to the large number of elementary events. (In the model, $256 = 4^4$).
 - ▶ Each V electron is quite rigidly sit in its V region. $p(\text{moving})=0.168$.
 - ▶ Rigid $\alpha + \beta$ tetrahedra give rise to the 4 α, β pairs.
 - ▶ Including the core, a 3-parameter model (C-V jumps are asymmetric) reproduces extremely well the distribution.



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- α probs. without core (HF).

p	V1	V2	V3	V4	Model	$p(\text{Model})$
0.4960	1	1	1	1	$s^4 + 6s^2j^2 + 8sj^3 + 9j^4$	0.4930
0.0377	2	0	1	1	$s^3j + 2s^2j^2 + 5sj^3 + 4j^4$	0.0374
0.0041	2	2	0	0	$2s^2j^2 + 2sj^3 + 2j^4$	0.0046
0.0021	3	1	0	0	$s^2j^2 + 2sj^3 + j^4$	0.0024
0.0000	4	0	0	0	j^4	0.0000

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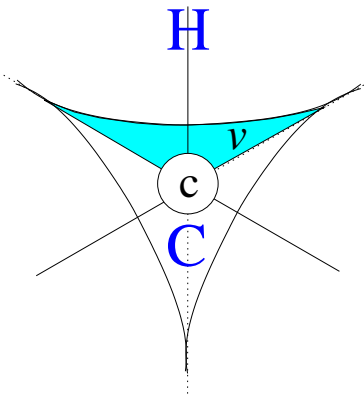
Is deeper insight possible? Joint atomic & localization regions analysis.

- What happens inside a localization basin? (e.g. $V(c,H)$)
 - ▶ Part of the bond valence is in C, part in H
 - ▶ $V(C,H)=V=H1 \cup v1$; $C=Carbon\ atom=c \cup v1 \cup v2 \cup v3$

- Populations: $\langle n \rangle_A$ & δ 's ($\delta \equiv -2cov$)

C	H	c	V	v
5.816	1.034	2.083	1.967	0.933
cv	cH	v1,v2	v1,H1	v1,H2
0.075	0.048	0.125	0.632	0.099

- $\delta(c,V) = 0.125 = 0.075 + 0.049$
- $\delta(V1,V2) = 0.371 = 0.043 + 2 \times 0.099 + 0.125$
- $\delta(C,H) = 0.977 = 0.048 + 3 \times 0.099 + 0.632$
- Rigid tetrahedra with mobile e 's within lobes.

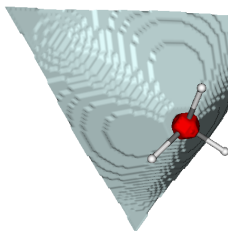


- For the isolated C atom, $\delta(c,v) \approx 0.10$, $\delta(v1,v2) \approx 0.37$

Are other localization regions possible? JCC 28, 442 (2007)

- Use p 's to build domains:

- ▶ Maximum probability domains (MPD's): Shape optimization $\max(p(n))$.
- ▶ Minimum variance domains (MVD's): Shape optimization $\min(\text{relvar})$



Core	$p(0)$	$p(1)$	$p(2)$	$p(3)$	$p(4)$
MPD	0.003	0.090	0.785	0.118	0.008
MVD	0.003	0.090	0.785	0.115	0.007
ELF	0.001	0.065	0.784	0.138	0.001

V(C,H)	$p(0)$	$p(1)$	$p(2)$	$p(3)$	$p(4)$
MPD	0.024	0.224	0.552	0.174	0.024
MVD	0.023	0.219	0.552	0.179	0.025
ELF	0.002	0.218	0.549	0.182	0.026

- ▶ In CH_4 , $p(2)$ MPD's very close to ELF basins (symmetry). MPD's & MVD's may overlap.

Conclusions, References

- More quantitative ELF analyses are possible and useful.
- Precise meaning of ELF EDFs?
- Even though ELF basin populations provide nice Lewis values in many cases... their statistical decomposition shows fairly wide distributions.
- Some papers:
 - ▶ *Chem. Phys. Lett.* **454**, 396 (2008);
 - ▶ *J. Phys. Chem. A* **111**, 1084 (2007);
 - ▶ *J. Chem. Phys.* **126**, 094102 (2007);
 - ▶ *Phys. Chem. Chem. Phys.* **9**, 1087 (2007);
 - ▶ *J. Chem. Phys.* **127**, 144103 (2007);
 - ▶ *J. Chem. Phys.* **131**, 124125 (2009);.
 - ▶ *Theor. Chem. Acc.*, DOI:10.1007/s00214-010-0809-4 (2010).